

## 7-3 Review and Reinforcement

### Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

| Number | Prefix |
|--------|--------|
| 1      | mono   |
| 2      | di     |
| 3      | tri    |
| 4      | tetra  |
| 5      | penta  |
| 6      | hexa   |
| 7      | hepta  |
| 8      | octa   |
| 9      | nona   |
| 10     | deca   |

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

cation

hydrates

electronegative

ionic

anion

anhydrides

hydrogen ions

molecular

11. Ionic compounds that absorb water into their solid structures form substances called

**—hydrate**

12. When naming an ionic substance, the name of the cation \_\_\_\_\_ comes first.

13. To name a binary molecular compound, add the suffix -ide to the more electronegative of the two elements.

14. By one definition, an acid is a molecular compound that dissolves in water to produce hydrogen ions

15. The name of an acid is usually derived from the name of its characteristic anion.

16. Prefixes are often used when naming molecular compounds.

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