

Chemistry Bingo

BUNDLE



Calculating Molar Mass

Instructions:
Calculate the molar mass for each of the compounds listed below. Write your answers in the space provided.



Example: H_2O

$(2 \times 1 \text{ atomic mass for } H) + (1 \times 16 \text{ atomic mass for } O) = \text{molar mass of } H_2O, 18 \text{ g/mol}$

1. CH_4	Formula: _____	Molar Mass: _____
2. CO_2	Formula: _____	Molar Mass: _____
3. $NaCl$	Formula: _____	Molar Mass: _____
4. H_2SO_4	Formula: _____	Molar Mass: _____



Percent Composition

Instructions:
Calculate the percent composition of each element in the compounds listed below. Write your answers in the space provided.



Example: H_2O

$\frac{2 \times 1 \text{ for } H}{18 \text{ for } H_2O} \times 100 = 11.1\% \text{ atomic mass for } H, 17.8\% \text{ for } O$

1. CH_4	Element: _____	%: _____
2. CO_2	Element: _____	%: _____
3. $NaCl$	Element: _____	%: _____
4. H_2SO_4	Element: _____	%: _____

**60
STUDENT
CARDS!**

**ANSWER
SHEET!**

Molar Mass				
C	H	N	O	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	

Percent Composition				
C	H	N	O	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	
12.01	1.01	14.01	16.00	

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