

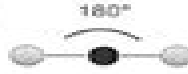
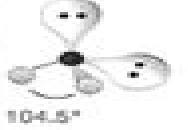
Molecular Geometry

How can molecular shapes be predicted using the VSEPR theory?

Why?

When you draw a Lewis structure for a molecule on paper, you are making a two-dimensional representation of the atoms. In reality however, molecules are not flat—they are *three-dimensional*. The true shape of a molecule is important because it determines many physical and chemical properties for the substance. In this activity you will learn how to predict molecular shapes.

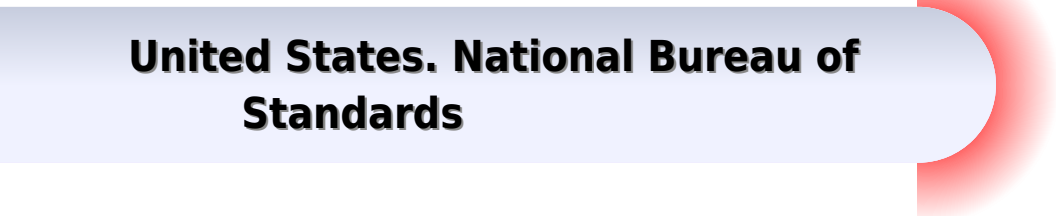
Model 1 – Lewis Structures

Lewis Structures			
1. H_2CO	$\begin{array}{c} \text{H} & & \text{O} \\ & \backslash & / \\ & \text{C} \\ & / & \backslash \\ \text{H} & & \text{H} \end{array}$	H_2CO 3 electron domains (3 bonding, 0 nonbonding)	 Trigonal Planar
2. BeF_2	$\begin{array}{c} \text{F} & & \text{F} \\ & \backslash & / \\ & \text{Be} \\ & / & \backslash \\ \text{F} & & \text{F} \end{array}$	BeF_2 2 electron domains (2 bonding, 0 nonbonding)	 Linear
3. CH_4	$\begin{array}{c} \text{H} \\ \\ \text{H} - \text{C} - \text{H} \\ \\ \text{H} \end{array}$	CH_4 4 electron domains (4 bonding, 0 nonbonding)	 Tetrahedral
4. NH_3	$\begin{array}{c} \text{H} \\ \\ \text{H} - \text{N} - \text{H} \\ \\ \cdot\cdot \end{array}$	NH_3 4 electron domains (3 bonding, 1 nonbonding)	 Pyramidal
5. H_2O	$\begin{array}{c} \cdot\cdot \\ \\ \text{O} - \text{H} \\ \\ \text{H} \end{array}$	H_2O 4 electron domains (2 bonding, 2 nonbonding)	 Bent
6. CO_2	$\begin{array}{c} \cdot\cdot & & \cdot\cdot \\ & \backslash & / \\ & \text{C} \\ & / & \backslash \\ \text{O} & & \text{O} \end{array}$	CO_2 2 electron domains (2 bonding, 0 nonbonding)	 Linear

Lone pair = $\cdot\cdot$

Molecular Geometry Key

**United States. National Bureau of
Standards**



Molecular Geometry Key:

The book delves into Molecular Geometry Key. Molecular Geometry Key is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Molecular Geometry Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Molecular Geometry Key
- Chapter 2: Essential Elements of Molecular Geometry Key
- Chapter 3: Molecular Geometry Key in Everyday Life
- Chapter 4: Molecular Geometry Key in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Molecular Geometry Key. The first chapter will explore what Molecular Geometry Key is, why Molecular Geometry Key is vital, and how to effectively learn about Molecular Geometry Key.
3. In chapter 2, this book will delve into the foundational concepts of Molecular Geometry Key. The second chapter will elucidate the essential principles that need to be understood to grasp Molecular Geometry Key in its entirety.
4. In chapter 3, the author will examine the practical applications of Molecular Geometry Key in daily life. This chapter will showcase real-world examples of how Molecular Geometry Key can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Molecular Geometry Key in specific contexts. This chapter will explore how Molecular Geometry Key is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Molecular Geometry Key. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Molecular Geometry Key.

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