

Unit: Stoichiometry Answer key "Mole-Mole Calculations" - WS #1

Directions: For each of the following problems, show your balanced chemical equation and all of your calculations.

 How many moles of potassium oxide (K₂O) will be formed when 1.52 moles of potassium reacts with oxygen according to the following reaction:

4 K + O₂ → 2 K₂O
1.52 mol K x
$$\frac{2 \mod K_2 O}{4 \mod K}$$
 = 0.760 mol K₂O

How many moles of copper are needed to react with sulfur to produce 0.25 moles of copper(I) sulfide according to the following reaction:

$$\begin{array}{cccc} \textbf{2 Cu} & + \textbf{S} & \rightarrow \textbf{Cu}_2\textbf{S} \\ 0.25 & mol & Cu_2Sx & \frac{2 \ mol \ Cu}{1 \ mol \ Cu_2S} = 0.50 \ mol \ Cu \end{array}$$

How many moles of sodium are needed to react with sulfuric acid to produce 3.75 moles of sodium sulfate according to the following (unbalanced) reaction:

2_ Na + _____ H₂(SO₄)
$$\Rightarrow$$
 _____ Na₂(SO₄) + _____ H₂
3.75 mol Na₂SO₄ \times $\frac{2 \text{ mol Na}}{1 \text{ mol Na}_2SO_4} = 7.50 \text{ mol Na}$

How many moles of calcium oxide will be produced when 1.6 moles of Iron(III)oxide reacts with calcium phosphate according to the following (unbalanced) reaction:

____ Fe₂O₃ + ____ Ca₃(PO₄)₂
$$\rightarrow$$
 ___ 2__ Fe(PO₄) + ___ 3__ CaO
1.6 mol Fe₂O₃ x $\frac{3 mol CaO}{1 mol Fe2O3 = 4.8 mol Ca}$

4. How many moles of oxygen are produced when 5.12 moles of magnesium chlorate decomposes into magnesium chloride and oxygen?

Word equation: magnesium chlorate → magnesium chloride + oxygen

Formula equation:
$$Mg(ClO_3)_2 \rightarrow MgCl_2 + 3O_2$$
 (must be balanced!)
$$5.12 \mod Mg(ClO_3)2 \times \frac{3 \mod O_2}{1 \mod Mg(ClO_3)2} = 15.4 \mod O_2$$

How many moles of Chromium(III) nitrate are produced when Chromium reacts with 0.85 moles of Lead(IV) nitrate to produce chromium(III) nitrate and lead?

Word equation: chromium + lead (IV) nitrate → chromium (III) nitrate + lead

Formula equation:
$$4\text{Cr} + 3\text{Pb}(\text{NO}_3)_4 \Rightarrow 4\text{Cr}(\text{NO}_3)_3 + 3\text{Pb}$$
 (must be balanced!)
$$0.85 \ mol \ Pb(NO_3)_4 \times \frac{4 \ mol \ Cr(NO_3)_3}{3 \ mol \ Pb(NO_3)_4} = 1.1 \ mol \ Cr(NO_3)_3$$

Mole 2 Answer Key With Work

Disha Experts

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The Practice of Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can t do chemistry if they can t do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book s unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http www bfwpub com pdfs wink POCPowerPoint Final ppt 832KB Spectrum Reading for Theme and Details in Literature Spectrum, Carson Dellosa Education, 2015-07-29 Reading words is just the first step help children comprehend the message by comparing and contrasting characters finding themes supporting answers with text and more using Spectrum R Focus Reading for Theme and Details in Literature for grade 5 Spectrum Focus takes aim at specific areas of study and helps children gain mastery by honing one skill at a time With skill specific instruction this standards based workbook elevates critical thinking through extensive introductions and explanations guided and independent practice comprehensive assessments and performance tasks it s the perfect resource to help children meet and exceed expectations **Homework Helpers: Chemistry** Greg Curran, 2025-09-12 Homework Helpers Chemistry is a user friendly review book that will make every student or parent trying to help their child feel like he or she has a private Chemistry tutor Concepts are explained in clear easy to understand language and problems are worked out with step by step methods that are easy to follow Each lesson comes with numerous review questions and answer keynotes that explain each correct answer and why it's correct This book covers all of the topics in a typical one year Chemistry curriculum including A systematic approach to problem solving conversions and the use of units Naming compounds writing formulas and balancing chemical equations Gas laws chemical kinetics acids and bases electrochemistry and more While Homework Helpers Chemistry is an excellent review for any standardized Chemistry test including the SAT II its real value is in providing support and guidance during the year s entire course of study 23 Year-wise JEE MAIN Chapter-wise Previous Year Solved Papers (2002 - 2024) 16th Edition | Physics, Chemistry & Mathematics PYOs Question Bank | Fully Solved | Disha Experts, The updated 16th Edition of 23 Years JEE Main Topic wise Solved Papers 2002 24 provides the past 11 years AIEEE 2002 12 Solved Papers and 12 years of JEE Main 2013 2024 Papers The book has been divided into 3 parts Physics Chemistry and Mathematics Each subject is further distributed into around 28 30 chapters each as per NCERT Thus making it 90 Chapters in all The book includes 1 paper of 2024 Ph 1 2023 Ph 1 2022 Ph 1 2021 Ph 1 February 2020 Ph 1 January 2 papers of 2019 1 of Ph I 1 of Phase II Each Chapter provides questions pertaining to all the concepts related to it from 2002 to 2023 Exams A total of 25 Question Papers including the AIEEE 2011 Rescheduled paper 2019 Ph II Paper have been distributed into these topics The questions in each Chapter are immediately

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