

Ch. 9 Practice Test

Chapter: Stoichiometry

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. A balanced chemical equation allows one to determine the
- mole ratio of any two substances in the reaction.
 - energy released in the reaction.
 - electron configuration of all elements in the reaction.
 - mechanism involved in the reaction.
- _____ 2. The coefficients in a chemical equation represent the
- masses, in grams, of all reactants and products.
 - relative numbers of moles of reactants and products.
 - number of atoms in each compound in a reaction.
 - number of valence electrons involved in the reaction.
- _____ 3. To determine the limiting reactant in a chemical reaction involving known masses of the two reactants, which of the following would be most useful?
- determining the masses of 100 mol of each reactant
 - determining the molar masses of the products
 - calculating bond energies
 - calculating the mass of a single product formed from each reactant
- _____ 4. How many mole ratios can be correctly obtained from the chemical equation $2\text{Al}_2\text{O}_3(l) \rightarrow 4\text{Al}(s) + 3\text{O}_2(g)$
- 3
 - 4
 - 6
 - 8

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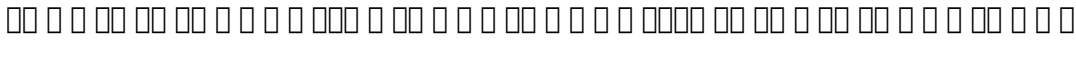
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