

PRE TEST

Chapter 6 Test-General/Honors Chemistry

Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

- | | | | |
|-------|--------------|---------------------------|--|
| C.2.2 | <u> a </u> | 1. cation | a. a positively charged ion |
| C.2.2 | <u> g </u> | 2. electronegativity | b. an element whose highest occupied <i>s</i> or <i>p</i> sublevels are partially filled |
| C.2.5 | <u> j </u> | 3. inner transition metal | c. the energy required to remove an electron from an atom |
| C.2.6 | <u> i </u> | 4. alkali earth metal | d. good conductor of heat and electric current |
| C.2.6 | <u> h </u> | 5. noble gas | e. any element whose highest occupied <i>s</i> sublevel and a nearby <i>d</i> sublevel contain electrons |
| C.2.5 | <u> b </u> | 6. representative element | f. any nonmetal in group 7A |
| C.2.2 | <u> d </u> | 7. metal | g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) |
| C.2.6 | <u> f </u> | 8. halogen | h. an element in which the highest occupied <i>s</i> and <i>p</i> sublevels are completely filled |
| C.2.5 | <u> e </u> | 9. transition metal | i. an element in group 2A |
| C.2.2 | <u> c </u> | 10. ionization energy | j. an element whose highest <i>s</i> sublevel and a nearby <i>f</i> sublevel contain electrons |

Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- | | | | | |
|-------|--------------|--------------------------------------|--------------------------|------------------------------|
| C.2.6 | <u> b </u> | 11. The alkali metals do not include | a. Li | c. Na |
| | | | b. Ca | d. Rb |
| C.2.6 | <u> c </u> | 12. The element iodine, I, is a | a. period 5 alkali metal | c. period 5 halogen |
| | | | b. period 4 halogen | d. period 5 transition metal |

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Forensic Chemistry Michael Grossman, 2021-12-20 FORENSIC CHEMISTRY FUNDAMENTALS strives to help scientists workplace health safety and other areas are included And two issues of Canadian legal process are argued as essays in the final two chapters *Modern Methods of Drug Discovery* Alexander Hillisch, Rolf Hilgenfeld, 2012-11-28 Research in the pharmaceutical industry today is in many respects quite different from what it used to be only fifteen years ago There have been dramatic changes in approaches for identifying new chemical entities with a desired biological activity While chemical modification of existing leads was the most important approach in the 1970s and 1980s high throughput screening and structure based design are now major players among a multitude of methods used in drug discovery Quite often companies favor one of these relatively new approaches over the other e.g screening over rational design or vice versa but we believe that an intelligent and concerted use of several or all methods currently available to drug discovery will be more successful in the medium term What has changed most significantly in the past few years is the time available for identifying new chemical entities Because of the high costs of drug discovery projects pressure for maximum success in the shortest possible time is higher than ever In addition the multidisciplinary character of the field is much more pronounced today than it used to be As a consequence researchers and project managers in the pharmaceutical industry should have a solid knowledge of the more important methods available to drug discovery because it is the rapidly and intelligently combined use of these which will determine the success or failure of preclinical projects Foundations of Inorganic Chemistry Gary Wulfsberg, 2017-11-02 Foundations of Inorganic Chemistry by Gary Wulfsberg is our newest entry into the field of Inorganic Chemistry textbooks designed uniquely for a one semester stand alone course or to be used in a full year inorganic sequence Foundations of Inorganic Chemistry by Gary Wulfsberg is our newest entry into the field of Inorganic Chemistry textbooks designed uniquely for a one semester stand alone course or to be used in a full year inorganic sequence By covering virtually every topic in the test from the 2016 ACS Exams Institute this book will prepare your students for success The new book combines careful pedagogy clear writing beautifully rendered two color art and solved examples with a broad array of original chapter ending exercises It assumes a background in General Chemistry but reviews key concepts and also assumes enrollment in a Foundations of Organic Chemistry course Symmetry and molecular orbital theory are introduced after the student has developed an understanding of fundamental trends in chemical properties and reactions across the periodic table which allows MO theory to be more broadly applied in subsequent chapters Use of this text is expected to increase student enrollment and build students appreciation of the central role of inorganic chemistry in any allied field Key Features Over 900 end of chapter exercises half answered in the back of the book Over 180 worked examples Optional experiments demos Clearly cited connections to other areas in chemistry and chemical sciences Chapter opening biographical vignettes of noted scientists in Inorganic Chemistry Optional General Chemistry review sections Originally rendered two color illustrations

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