

Types of Chemical Reactions Worksheet #1

H/Chemistry

For each of the following equations, identify what kind of reaction it represents: double replacement, single replacement, decomposition, or synthesis (composition).

- 1) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ **Synthesis**
- 2) $\text{CaI}_2 + \text{Cl}_2 \rightarrow \text{CaCl}_2 + \text{I}_2$ **Single Replacement**
- 3) $3\text{KOH} + \text{AlCl}_3 \rightarrow \text{Al(OH)}_3 + 3\text{KCl}$ **Double Replacement**
- 4) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ **Synthesis**
- *5) $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ **Double Replacement**
- 6) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ **Decomposition**
- 7) $3\text{CuSO}_4 + 2\text{Al} \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{Cu}$ **Single Replacement**
- 8) $\text{Na}_2\text{S} + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3 + \text{Ag}_2\text{S}$ **Double Replacement**
- 9) $3\text{H}_2 + \text{N}_2 \rightarrow 2\text{NH}_3$ **Synthesis**
- *10) $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$ **Single Replacement**

*Remember: "H₂O" can also be thought of as "HOH"

Complete each of the following equations as needed to make it the type of reaction indicated. Be sure to write each formula correctly.

- 11) Double replacement: $\text{Na}_2\text{CrO}_4 + \text{PbCl}_2 \rightarrow 2\text{NaCl} + \text{PbCrO}_4$
- 12) Single replacement: $\text{Cl}_2 + 2\text{NaBr} \rightarrow 2\text{NaCl} + \text{Br}_2$
- 13) Decomposition: $\text{Mg(ClO}_3)_2 \rightarrow \text{MgCl}_2 + 3\text{O}_2$
- 14) Synthesis: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- 15) Double replacement: $3\text{Ca(OH)}_2 + 2\text{FeCl}_3 \rightarrow 2\text{Fe(OH)}_3 + 3\text{CaCl}_2$
- 16) Single replacement: $\text{Fe} + \text{Cu(NO}_3)_2 \rightarrow \text{Fe(NO}_3)_2 + \text{Cu}$ [Assume Fe²⁺]
- 17) Decomposition: $2\text{Hg}_2\text{O} \rightarrow 4\text{Hg} + \text{O}_2$
- 18) Synthesis: $\text{S} + \text{O}_2 \rightarrow \text{SO}_2$
- 19) Double replacement: $\text{AgNO}_3 + \text{KI} \rightarrow \text{AgI} + \text{KNO}_3$
- 20) Single replacement: $\text{Cu} + 2\text{AgNO}_3 \rightarrow 2\text{Ag} + \text{Cu(NO}_3)_2$ [Copper (II) is used here]

Modern Chemistry Chemical Reactions Answer Key

Dr. Lassar-Cohn



Modern Chemistry Chemical Reactions Answer Key:

Principles of Modern Chemistry David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom Part I. -

An Introduction to Modern Inorganic Chemistry John Newton Friend, 1914 **Journal of the Photographic Society of**

London, 1905 Vols for 1853 include the transactions of the Royal Photographic Society of Great Britain *Modern Electrochemistry* 2A John O'M. Bockris, Amulya K.N. Reddy, Maria E. Gamboa-Aldeco, 2007-05-08 This book had its nucleus in some lectures given by one of us J O M B in a course on electrochemistry to students of energy conversion at the University of Pennsylvania It was there that he met a number of people trained in chemistry physics biology metallurgy and materials science all of whom wanted to know something about electrochemistry The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds was thereby engendered The lectures were recorded and written up by Dr Klaus Muller as a 293 page manuscript At a later stage A K N R joined the effort it was decided to make a fresh start and to write a much more comprehensive text Of methods for direct energy conversion the electrochemical one is the most advanced and seems the most likely to become of considerable practical importance Thus conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met Corrosion is recognized as having an electrochemical basis The synthesis of nylon now contains an important electrochemical stage Some central biological mechanisms have been shown to take place by means of electrochemical reactions A number of American organizations have recently recommended greatly increased activity in training and research in electrochemistry at universities in the United States **Volume 1: Modern Electrochemistry** John O'M. Bockris, Amulya K.N.

Reddy, 1998-06-30 This book had its nucleus in some lectures given by one of us J O M B in a course on electrochemistry to students of energy conversion at the University of Pennsylvania It was there that he met a number of people trained in chemistry physics biology metallurgy and materials science all of whom wanted to know something about electrochemistry The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds

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Modern Physical Organic Chemistry Eric V. Anslyn, Dennis A. Dougherty, 2006 Making explicit the connections between physical organic chemistry and critical fields such as organometallic chemistry materials chemistry bioorganic chemistry and biochemistry this book escorts the reader into an area that has been thoroughly updated in recent times

Modern Experimental Chemistry George W. Jr. Latimer, 2012-12-02 Modern Experimental Chemistry provides techniques of qualitative analysis that reinforce experiments on ionic equilibria This book includes the determination of water in hydrated salts identification of an organic compound after determining its molecular weight and nonaqueous titration of a salt of a weak acid The calculation of chemical stoichiometry calculation of thermodynamic properties by determining the change in equilibrium with temperature and chromium chemistry are also covered This compilation contains enough experiments for classes which have six hours of laboratory two 3 hour meetings per week to last two semesters This publication is intended for chemistry students as an introductory manual to chemistry laboratory

Women In Their Element: Selected Women's Contributions To The Periodic System Annette Lykknes, Brigitte Van Tiggelen, 2019-08-05 2019 celebrated the 150th anniversary of Mendeleev's first publication of the Periodic Table of Chemical Elements This book offers an original viewpoint on the history of the Periodic Table a collective volume with short illustrated papers on women and their contribution to the building and the understanding of the Periodic Table and of the elements themselves Few existing texts deal with women's contributions to the Periodic Table A book on women's work not only helps make historical women chemists more visible it also sheds light on the multifaceted character of the work on the chemical elements and their periodic relationships Stories of female input contribute to the understanding of the nature of science of collaboration as opposed to the traditional depiction of the lone genius While the discovery of elements is a natural part of this collective work the book goes beyond discovery histories Stories of women contributors to the chemistry of the elements also include understanding the concept of element identifying properties developing analytical methods mapping the radioactive series finding applications of elements and the participation of women as audiences when new elements were presented at lectures The book contains chapters on pre

periodic table contributions as well as recent discoveries unknown stories as well as more famous ones with an emphasis on work conducted in the late 19th century and early 20th century Elements from different groups in the periodic table are included so as to represent a variety of chemical contexts Fundamentals of Chemistry: A Modern Introduction (1966)

Frank Brescia, 2012-12-02 Fundamentals of Chemistry A Modern Introduction focuses on the formulas processes and methodologies used in the study of chemistry The book first looks at general and historical remarks definitions of chemical terms and the classification of matter and states of aggregation The text then discusses gases Ideal gases pressure of a gas confined by a liquid Avogadro's Law and Graham's Law are described The book also discusses aggregated states of matter atoms and molecules chemical equations and arithmetic thermochemistry and chemical periodicity The text also highlights the electronic structures of atoms Quantization of electricity spectra of elements quantization of the energy of an electron associated with nucleus the Rutherford Bohr nuclear theory hydrogen atom and representation of the shapes of atomic orbitals are explained The text also highlights the types of chemical bonds hydrocarbons and their derivatives intermolecular forces solutions and chemical equilibrium The book focuses as well on ionic solutions galvanic cells and acids and bases It also discusses the structure and basicity of hydrides and oxides The reactivity of hydrides charge of dispersal and basicity effect of anionic charge inductive effect and basicity and preparation of acids are described The book is a good source of information for readers wanting to study chemistry Modern Chemistry Nicholas D. Tzimopoulos, 1990 **An**

Introduction to Modern Scientific Chemistry Dr. Lassar-Cohn, 1901 *Investigations on the Theory of the Photographic Process* Samuel Edward Sheppard, Charles Edward Kenneth Mees, 1907 Educart CBSE Question Bank Class 9 Science 2024-25 (For 2025 Board Exams) Educart, 2024-06-17 What You Get Time Management Charts Self evaluation

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Operation Manual for Spaceship Earth Synergy is the behavior of whole systems unpredicted by separately observed behaviors of any of the system's separate parts In a similar vein one might define an intellectual synergy as an improvement in our understanding of the behavior of a system unpredicted by separately acquired viewpoints of the activities of such a system Such considerations underlie and provide a motivation for an interdisciplinary approach to the problem of unraveling the deeper mysteries of cellular metabolism and organization and have led a number of pioneering spirits many represented

in the pages which follow to consider biological systems from an electrochemical standpoint is itself of course an interdisciplinary branch of Now electrochemistry science and there is no doubt that many were introduced to it via Bockris and Reddy's outstanding wide ranging and celebrated textbook Modern Electrochemistry If I am to stick my neck out and seek to define bioelectrochemistry I would take it to refer to the study of the mutual interactions of electrical fields and biological materials including living systems An Introduction to Modern Scientific Chemistry in the Form of Popular Lectures Suited for University Extension Students and General Readers Dr. Lassar-Cohn,1901 Science ,2009-10-05 The complete illustrated science encyclopedia covering the history key discoveries inventions and people This remarkable reference book reveals the story of scientific progress from the invention of the wheel to 21st century climate solutions including everything from ancient Greek geometry and quantum physics to the worldwide web Explore every key moment of scientific discovery and find out how the concepts inventions and the individuals behind them have changed our world With stunning artworks and authoritative information this makes even complex scientific subjects easily comprehensible

Modern Electrochemistry John Bockris,2012-12-06 This book had its nucleus in some lectures given by one of us J O M B in a course on electrochemistry to students of energy conversion at the University of Pennsylvania It was there that he met a number of people trained in chemistry physics biology metallurgy and materials science all of whom wanted to know something about electrochemistry The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds was thereby engendered The lectures were recorded and written up by Dr Klaus Muller as a 293 page manuscript At a later stage A K N R joined the effort it was decided to make a fresh start and to write a much more comprehensive text Of methods for direct energy conversion the electrochemical one is the most advanced and seems the most likely to become of considerable practical importance Thus conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met Corrosion is recognized as having an electrochemical basis The synthesis of nylon now contains an important electrochemical stage Some central biological mechanisms have been shown to take place by means of electrochemical reactions A number of American organizations have recently recommended greatly increased activity in training and research in electrochemistry at universities in the United States **Lab Experiments for Modern Chemistry** Tzimopoulo,1990 **Liquids, Solutions, and Interfaces** W. Ronald Fawcett,2004-07 Fawcett chemistry University of California Davis introduces modern topics in solution chemistry to senior undergraduates and graduate students who have completed two semesters or three quarters of chemical thermodynamics and statistical mechanics The Montreal Medical Journal George Edgeworth Fenwick,Thomas George Roddick,George Ross,1906

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