

The goals of the Initial Preventive Physical Examination (IPPE), also known as the "Welcome to Medicare Visit," are health promotion and disease detection. This document explains the components included in the IPPE. All components of the IPPE must be provided, or provided and referred, prior to submitting a claim for the IPPE.

## Components of the IPPE

ACQUIRE BENEFICIARY HISTORY	ELEMENTS
<input type="checkbox"/> 1. Review of beneficiary's medical and social history	<p>At a minimum, obtain the following:</p> <ul style="list-style-type: none"> <li>• Past medical/surgical history (experiences with illnesses, hospital stays, operations, allergies, injuries, and treatments);</li> <li>• Current medications and supplements (including calcium and vitamins);</li> <li>• Family history (review of medical events in the family, including diseases that may be hereditary or place the beneficiary at risk);</li> <li>• History of alcohol, tobacco, and illicit drug use;</li> <li>• Diet; and</li> <li>• Physical activities.</li> </ul>
<input type="checkbox"/> 2. Review of beneficiary's potential risk factors for depression and other mood disorders	<p>Use any appropriate screening instrument for persons without a current diagnosis of depression recognized by national professional medical organizations to obtain current or past experiences with depression or other mood disorders.</p>
<input type="checkbox"/> 3. Review of beneficiary's functional ability and level of safety	<p>Use any appropriate screening questions or standardized questionnaire recognized by national professional medical organizations to review, at a minimum, the following areas:</p> <ul style="list-style-type: none"> <li>• Hearing impairment;</li> <li>• Activities of daily living;</li> <li>• Falls risk; and</li> <li>• Home safety.</li> </ul>
BEGIN EXAMINATION	ELEMENTS
<input type="checkbox"/> 4. An examination	<p>Obtain the following:</p> <ul style="list-style-type: none"> <li>• Height, weight, and blood pressure;</li> <li>• Visual acuity screen;</li> <li>• Measurement of body mass index; and</li> <li>• Other factors deemed appropriate based on the beneficiary's medical and social history and current clinical standards.</li> </ul>
<input type="checkbox"/> 5. End-of-life planning	<p>End-of-life planning is a required service, upon the beneficiary's consent. End-of-life planning is verbal or written information provided to the beneficiary regarding:</p> <ul style="list-style-type: none"> <li>• The beneficiary's ability to prepare an advance directive in the case that an injury or illness causes the beneficiary to be unable to make health-care decisions; and</li> <li>• Whether or not the physician is willing to follow the beneficiary's wishes as expressed in the advance directive.</li> </ul>
COUNSEL BENEFICIARY	ELEMENTS
<input type="checkbox"/> 6. Education, counseling, and referral based on the previous five components	<p>Based on the results of the review and evaluation services provided in the previous five components, provide education, counseling, and referral as appropriate.</p>
<input type="checkbox"/> 7. Education, counseling, and referral for other preventive services	<p>Complete a brief written plan, such as a checklist, to be given to the beneficiary for obtaining a screening electrocardiogram (ECG), as appropriate, and the appropriate screenings and other preventive services that are covered as separate Medicare Part B benefits. (See below for a list of Medicare-covered preventive services.)</p>

# Medicare Preventive Service Quick Reference Guide 2013

**C Cleary**



## **Medicare Preventive Service Quick Reference Guide 2013:**

## Reviewing **Medicare Preventive Service Quick Reference Guide 2013**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Medicare Preventive Service Quick Reference Guide 2013**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical  $\times 100$  what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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