

MBA Business Research Categories

Cont'd..

- Business Theory - Covers topics ranging from Why People Work to the characteristics of an entrepreneur.
- Career Development - Studies the issues and methods involved in career development, as well as the personal and organizational issues and practices relevant to individual careers.
- Case Studies - Suggestions of companies to do case studies on. Also identifies the Central Problems and Questions for a topic dealing with a case study.
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,1960 **Corporate Social Responsibility** John O. Okpara, Samuel O. Idowu, 2013-12-02 In today's global business environment it is no longer acceptable that a corporation does well simply by doing good It is expected With increasing pressures from stakeholders to improve the bottom line as well as to be good corporate citizens business leaders face tough decisions What social issues should we support What initiatives should we develop that will do the most good for the company as well as the cause Do we include social messages in our advertising encourage our employees to volunteer do we modify our business practices How do we integrate a new initiative into current strategies These and other challenges will continue to face future leaders This book provides thoughtful answers to these important questions and to many more The book offers suggestions on how to choose among major worthy causes and also how to measure the amount of good achieved both for the recipients and the companies themselves Of course all is not only about challenges there are loads of opportunities that go along with them but it's only responsible and sustainable leaders who would be able to spot these opportunities That is the future which awaits 21st century leaders *Research, Teaching and Learning in Higher Education*

Sally Brown, Brenda Smith, 2013-09-13 Exploring the issue of how educational staff can balance successfully their research and teaching activities this volume argues that the entire system governing the relationship amongst research teaching and learning should be dismantled and rebuilt focusing on symbiosis rather than conflict *The Work System Method* Steven Alter, 2006 The Work System Method is an organized approach that every organization can use for Recognizing that systems involve much more than IT Describing and understanding systems from a business viewpoint Analyzing and improving systems Improving communication between business and IT professionals Increasing the likelihood of successful

implementation Understanding the role and limitations of IT **Advanced Issues in the Economics of Emerging Markets** William A. Barnett, Bruno S. Sergi, 2020-06-09 Volume 27 of the International Symposia in Economic Theory and Econometrics series collects a range of unique and diverse chapters each investigating different spheres of development in emerging markets with a specific focus on significant engines of growth and advancement in the Asia Pacific economies

2012-2013 UNCG Graduate School Bulletin , Research on Project, Programme and Portfolio Management Roxana Cuevas, Constanta-Nicoleta Bodea, Pablo Torres-Lima, 2020-12-17 This book collects recent work presented at the 31st IPMA Congress which was held in Merida Mexico from September 30th to October 2nd 2019 It covers a range of project programme and portfolio management contexts with the general aim of integrating sustainability into project management The book is structured into three parts The first part covers concepts and approaches related to the integration of sustainability in project management The second part presents research on integrating sustainability into project management in different industries and regions The final part takes specific perspectives on integrating sustainability into project management related to learning and continuing competence development The book offers a valuable resource for all researchers interested in studying the emerging trends in incorporating sustainability in project programme and portfolio management **CAT | COMMON ADMISSION TEST| 10 YEARS'| CHAPTERWISE & TOPICWISE| SOLVED PAPERS|QUESTION BANK|2006- 2008|2017 - 2023 (Subject Name - VARC, DILR & QA)** Oswaal Editorial

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web jan 26 2016 the solubility product or ksp is a way of showing how effective a chemical is at dissolving into an aqueous solution the higher the ksp the more the chemical will dissolve into that solvent for this lab calcium hydroxide is created then ionized Ca(OH)_2 when calcium is put in water reacts as follows $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$

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web k sp of calcium hydroxide 1 author j m mccormick last update ocober 13 2013 introduction an equilibrium constant k is

related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined with eqn 2 to determine ΔH and ΔS from the temperature dependence of K

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web in today's experiment you will determine the solubility product ksp of calcium hydroxide Ca(OH)_2 by measuring the concentration of Ca(OH)_2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of Ca(OH)_2 is represented in equation form as shown below $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a pH regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ $5 \times 10^{-2} \text{ M}$ calcium ions $5 \times 10^{-2} \text{ M}$ hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$

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web the chemical equation used was $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ the solubility product was determined to be ksp Ca(OH)_2 from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

18.1 solubility product constant ksp chemistry libretexts - Feb 01 2023

web jul 12 2023 because the concentration of a pure solid such as $\text{Ca}_3(\text{PO}_4)_2$ is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $\text{Ca}_3(\text{PO}_4)_2 \rightleftharpoons 3\text{Ca}^{2+} + 2\text{PO}_4^{3-}$ $K_{\text{sp}} = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]^2$

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web lab report 8 determining of calcium hydroxide the purpose of this was to titrate a saturated Ca(OH)_2 solution with a standard hcl solution and find the ksp or the oh concentration was found and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochloric

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web question determining the ksp of calciumhydroxide $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ for this experiment the titration of calciumhydroxide with hydrochloric acid was investigated determining the ksp of calcium hydroxide $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ for this experiment the titration of calcium

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web question chem 182 experiment 8 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of Ca(OH)_2 is represented in equation form lution of calcium on a sparingly soluble ionic solid that is sparingly soluble in water a saturated aqueous as shown below

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for Ca(OH)_2 was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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web calculate the ksp for calcium hydroxide $\text{ksp Ca(OH)}_2 = 5.0 \times 10^{-6}$ find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for ksp for sodium hydroxide is 5×10^{-6}