Name: ___ Date: Mods:

Honors Chemistry Midterm: Review Packet

PART 1

Write in the number of significant figures for each of the following measurements:

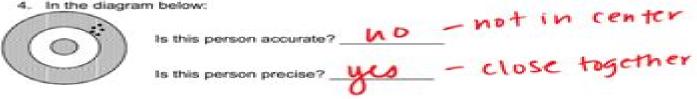
3 0.00290 g 4 50.00 m 2 0.070 kg /

- Using the following masses: 0.0562 kg, 124.213 g and 1635 mg.
 The answer should be reported as: ______g The answer should be reported as: ______g 1.635g
- 3. a) A 5.75 mL sample of mercury has a measured mass of 77.05 g. The density is 13.49/mL

b) Mercury's accepted density is 13.53 g/mL. The % error in your measurement is: . 961 70

acture ×100 113.639/ms -13.49/ms ×100

In the diagram below:



- 5. Identify the following as a chemical change (CC), physical change (PC), chemical property (CP), or physical property (PP):
 - a) burning wood CC

b) mass of wood PP

c) flammability of wood CP d) chopping up wood PC

Honors Chemistry Semester Review Packet Answers

Douglas W. Henderson

Honors Chemistry Semester Review Packet Answers:

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