F324-unofficial mark scheme (in progress)

Marks accounted for: 60/60

- Give the systematic name of the alcohol that is formed by this hydrolysis. (1)
 - Propan(e)-1,2,3-triol
- What are the other monomers that make up this polymer? (3)

(Hydrolysed by hot NaOH)

- Two different carboxylic acids with COO'(Na") The monomers have COOH
 as it's not referring to the hyrolysed products but the monomer units of
 the polymer I'm pretty sure that the question said: "What are the other
 products of this hydrolysis" or something along those lines
- One salt had a c=c double bond
- Why are people who drink a lot of goat's milk likely to suffer from CHD? (2)
 - Fatty acids build up in the arteries and lead to CHD?
 - Has trans isomer due to c=c double bond fatty acid linked to LDL's and hence Atherosclerosis
 - Trans-fatty acid in triglyceride of milk increases 'bad' cholesterol levels in blood, resulting in increased risk of coronary heart disease and stroke (straight from the spec)
 - trans fatty acids can pack more tightly and hence have higher melting points than cis fatty acids. --> so more likely to deposit on artery walls
 - The other two molecules were fully saturated fatty acids which would also lead to the formation of atheroma
- State how 4-aminophenol can behave as a base. (1)
- The lone pair on the Nitrogen atom in the amine group can accept a proton, forming a dative covalent bond with it to give NH3+.
- NH2 + H+ ---> NH3+
- 4-Aminophenol is produced by the reduction of 4-nitrophenol. Write an equation to show the production of 4-aminophenol from 4-nitrophenol. Use [H] to represent the reducing agent. (1)
- 4-Nitrophenol + 6[H] = 4 Aminophenol + 2H20
- 4-nitrophenol can be produced from 4-bromophenol. (4)
 - Complete the mechanism for this reaction.
 - Use "NO, as the electrophile. Include any intermediate and the products.
 - In the mechanism for this reaction, NO₂ substitutes for Br on the ring.

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M Carnoy

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Table of Contents F324 June 2014 Unofficial Mark Scheme

- 1. Understanding the eBook F324 June 2014 Unofficial Mark Scheme
 - The Rise of Digital Reading F324 June 2014 Unofficial Mark Scheme
 - Advantages of eBooks Over Traditional Books
- 2. Identifying F324 June 2014 Unofficial Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - o Features to Look for in an F324 June 2014 Unofficial Mark Scheme
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from F324 June 2014 Unofficial Mark Scheme
 - Personalized Recommendations

- F324 June 2014 Unofficial Mark Scheme User Reviews and Ratings
- F324 June 2014 Unofficial Mark Scheme and Bestseller Lists
- 5. Accessing F324 June 2014 Unofficial Mark Scheme Free and Paid eBooks
 - F324 June 2014 Unofficial Mark Scheme Public Domain eBooks
 - F324 June 2014 Unofficial Mark Scheme eBook Subscription Services
 - F324 June 2014 Unofficial Mark Scheme Budget-Friendly Options
- 6. Navigating F324 June 2014 Unofficial Mark Scheme eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - F324 June 2014 Unofficial Mark Scheme Compatibility with Devices
 - F324 June 2014 Unofficial Mark Scheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of F324 June 2014 Unofficial Mark Scheme
 - o Highlighting and Note-Taking F324 June 2014 Unofficial Mark Scheme
 - Interactive Elements F324 June 2014 Unofficial Mark Scheme
- 8. Staying Engaged with F324 June 2014 Unofficial Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers F324 June 2014 Unofficial Mark Scheme
- 9. Balancing eBooks and Physical Books F324 June 2014 Unofficial Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection F324 June 2014 Unofficial Mark Scheme
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine F324 June 2014 Unofficial Mark Scheme
 - Setting Reading Goals F324 June 2014 Unofficial Mark Scheme
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of F324 June 2014 Unofficial Mark Scheme
 - Fact-Checking eBook Content of F324 June 2014 Unofficial Mark Scheme

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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