



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the steps for naming the compound)



Steps for Naming:

1. Identify the cation (metal) and anion (non-metal).
2. Name the cation first, followed by the anion.
3. Use the appropriate suffix for the anion (usually -ide).
4. If the cation is a transition metal, include its oxidation state in parentheses.
5. If the anion is a polyatomic ion, use its name.
6. If the compound is a binary ionic compound, use the prefix system for the anion.

Binary Compounds

Formula	Name
NaCl	Sodium Chloride
CaO	Calcium Oxide
Fe ₂ O ₃	Iron(III) Oxide
Al ₂ Cl ₆	Aluminum Chloride
CO ₂	Carbon Dioxide
SO ₂	Sulfur Dioxide
NO ₂	Nitrogen Dioxide
H ₂ O	Water
CH ₄	Methane
NH ₃	Nitrogen
HCl	Hydrogen Chloride

Common Ions and Polyatomic Ions

Formula	Name
Na ⁺	Sodium
Ca ²⁺	Calcium
Fe ²⁺	Iron(II)
Fe ³⁺	Iron(III)
Al ³⁺	Aluminum
CO ₃ ²⁻	Carbonate
SO ₄ ²⁻	Sulfate
NO ₃ ⁻	Nitrate
H ₂ PO ₄ ⁻	Dihydrogen Phosphate
HPO ₄ ²⁻	Hydrogen Phosphate
PO ₄ ³⁻	Phosphate

Naming Compounds with a Central and Peripheral



Carbon Dioxide

Steps for Naming:

1. Identify the central atom and the peripheral atoms.
2. Name the central atom first, followed by the peripheral atoms.
3. Use the appropriate prefix for the peripheral atoms.
4. If the compound is a binary molecular compound, use the prefix system for the anion.

Naming Rules:

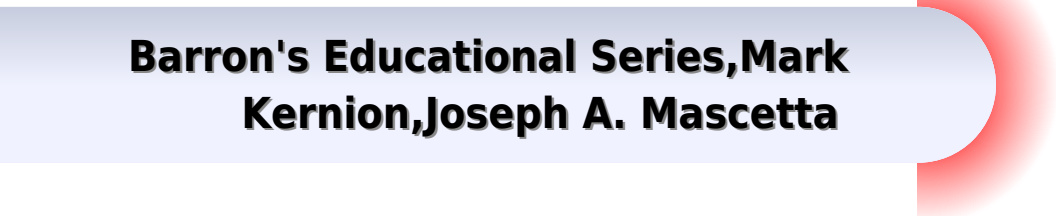
Formula	Name	Formula	Name
CO ₂	Carbon Dioxide	SO ₂	Sulfur Dioxide
CO	Carbon Monoxide	SO	Sulfur Monoxide
NO ₂	Nitrogen Dioxide	NO	Nitrogen Monoxide
NO	Nitrogen Monoxide	NO ₂	Nitrogen Dioxide
CO ₂	Carbon Dioxide	SO ₂	Sulfur Dioxide
CO	Carbon Monoxide	SO	Sulfur Monoxide
NO ₂	Nitrogen Dioxide	NO	Nitrogen Monoxide
NO	Nitrogen Monoxide	NO ₂	Nitrogen Dioxide

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
NaCl	Salt	NaOH	Sodium Hydroxide
H ₂ O	Water	CH ₄	Methane
HCl	Hydrogen Chloride	CO ₂	Carbon Dioxide
SO ₂	Sulfur Dioxide	NO ₂	Nitrogen Dioxide

Naming Compounds Study Guide

**Barron's Educational Series, Mark
Kernion, Joseph A. Mascetta**



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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The

remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

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Basic Concepts of Chemistry, Study Guide Leo J. Malone, 1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples

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