

Name _____

Molecular Geometries Lab Report Sheet

Formula	Lewis Structure	No of electron domains/lig	Electron domain geometry	VSEPR Molecular geometry	Hybridization of central atom	Polar ?
CO_2	$\text{O}=\text{C}=\text{O}$ 24 valence 24 2 charge clouds	0	Linear 180°	Linear 	sp	NP
BF_3	$\text{F}-\text{B}-\text{F}$ 24 32 3 charge clouds	0	Trigonal planar 120°		sp ²	NP
SnCl_2	$\text{Cl}-\text{Sn}-\text{Cl}$ 16 24	3	Bent <120°		sp ²	P
BF_4^-	$\left[\text{F}-\text{B}-\text{F} \right]^-$ 24 40	0	Tetrahedral 109.5°		sp ³	P
AsH_3	$\text{H}-\text{As}-\text{H}$ 8	4	Trigonal pyramidal 109.5°		sp ³	P
H_2S	$\text{H}-\text{S}-\text{H}$ 8	2	Bent 109.5°		sp ³	NP
NH_2^-	$\left[\text{H}-\text{N}-\text{H} \right]^-$ 8	2	Bent 109.5°		sp ³	P
POF_3	$\text{F}-\text{P}-\text{Cl}$ 10 24	0	Trigonal bipyre 120° + 90°		sp ³ d	P

Molecular Geometry Lab Answers

JA Banks



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Lab Manual for General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit www.whfreeman.com/gob

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Petrucci, William Harwood, Geoffrey Herring, 2001 Conformational Concept For Synthetic Chemist's Use: Principles And In Lab Exploitation Anatoly M Belostotskii, 2015-09-17 This innovative book presents an original account of the principles of conformational theory It has a strong focus on computational methodologies for conformational space exploration By revisiting basic conformational conventions considering experimental results which are often misinterpreted by organic chemists and qualitatively analyzing the potential energy surface the book helps non experts to understand molecular flexibility at the level required in contemporary research The book shows synthetic organic chemists how to perform successful conformational studies using widespread calculation packages click computational chemistry instead of being misguided by textbook based conformational analysis The monograph actually offers to synthetic chemists a new research tool that can significantly upgrade their ability to predict or at least explain regioselectivity and stereoselectivity in their own reactions **Integrated Approach to Coordination Chemistry** Rosemary A. Marusak, Kate Doan, Scott D.

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