## Stoichiometry Limiting Reactant

#### Answers

How many grams of calcium phosphate can be produced from the reaction between 2.50 L of 0.250 M calcium chloride with an excess of phosphoric acid?

2 How many milliliters of 1.50 M nitric acid is required to react with 100.0 g of cuprous oxide?

(3) Consider the following unbalanced reaction:

Balance the reaction. 3.25 g of NH, is allowed to react with 3.5 g of O2.

Which reactant is the limiting reagent?

O, is the limiting reagent.

b How many grams of NO are formed?

2.63 g of NO are formed.

E How much excess reagent remains after the reaction?

$$3.5 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{32 \text{ g O}_2} \times \frac{4 \text{ mol NH}_3}{5 \text{ mol O}_2} \times \frac{17.03 \text{ g NH}_3}{1 \text{ mol NH}_3} = 1.49 \text{ g NH}_3 \text{ used}$$

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# **Molarity Limiting Reactant Worksheet**

Michael Rajnik

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Principles of Chemical Engineering Processes Nayef Ghasem, Redhouane Henda, 2014-11-10 This book introduces the basic principles and calculation techniques used in chemical engineering It discusses problems in material and energy balances related to chemical reactors explains the concepts of dimensions units psychrometry steam properties and conservation of mass and energy and demonstrates how MATLAB and Simulink can be used to solve complicated problems This Second Edition contains additional homework problems and a new chapter related to single and multiphase systems Educational software downloadable exercises and a solutions manual are available with qualifying course adoption ,

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