

Chemical Bonding

Section 4: Ionic Bonding and Ionic Compounds

In this space, provide your answer to each of the items on physics that have completed most important or least answers each question.

1. The most representative ion represented by a transition element is a(n) _____ ion.
 - a. d-block transition metals
 - b. s-block transition metals
 - c. f-block transition metals
 - d. p-block transition metals
2. Compared with the energy of ionization, the energy of a(n) _____ ion is _____ energy.
 - a. d-block transition metals
 - b. s-block transition metals
 - c. f-block transition metals
 - d. p-block transition metals
3. The most representative ion represented by a transition element is a(n) _____ ion.
 - a. d-block transition metals
 - b. s-block transition metals
 - c. f-block transition metals
 - d. p-block transition metals
4. Compared with the energy of ionization, the energy of a(n) _____ ion is _____ energy.
 - a. d-block transition metals
 - b. s-block transition metals
 - c. f-block transition metals
 - d. p-block transition metals
5. The most representative ion represented by a transition element is a(n) _____ ion.
 - a. d-block transition metals
 - b. s-block transition metals
 - c. f-block transition metals
 - d. p-block transition metals
6. Compared with the energy of ionization, the energy of a(n) _____ ion is _____ energy.
 - a. d-block transition metals
 - b. s-block transition metals
 - c. f-block transition metals
 - d. p-block transition metals
7. The most representative ion represented by a transition element is a(n) _____ ion.
 - a. d-block transition metals
 - b. s-block transition metals
 - c. f-block transition metals
 - d. p-block transition metals

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