Chapter 3: Matter and Energy - Study Guide

*Do not forget about the extra resources available to help you study and prepare for this test!

- Textbook pages 81-93 (answers to odd questions in the back of the book)
- Mastering Chemistry Study Area: http://www.pearsonmylabandmastering.com/northamerica/
- Miss Marnik's Website: http://www.northallegheny.org/Page/16665
- Labs, problem packets, and homework assignments

Section 1 - Classifying Matter: (text pages 57-62)

Need to be able to:

- Define matter, atoms, molecules, element, compound, and mixture.
- Classify matter as and distinguish between solid, liquid, or gas.
- Classify matter as element, compound, or mixture.

Practice Problems:

- Match the following terms with the phrases below:
 - State of matter in which atoms or molecules pack close to each other in fixed locations
 - State of matter in which atoms or molecules are packed close together but are free to move around and by each other
 - c. Substance composed of only one type of atom or molecule
 - Matter composed of two or more different types of atoms or molecules combined in variable proportions
 - Mixture that has two or more regions with different compositions
 - f. Mixture that has the same composition throughout
- 2. Which of the following can be classified as matter?
 - a. atoms c. wind

e. potential energy

b. gas

d. liquids

f. temperature

g. carbon h. light

heterogeneous mixture

homogeneous mixture

liquid

solid

mixture.

pure substance.

- Classify each type of matter as a pure substance or a mixture. If it is a pure substance, classify it as an
 element or a compound. If it is a mixture, classify it as homogeneous or heterogeneous.
 - a. limestone (CaCo₁)

e. Chex mix

b. orange juice with pulp

f. Table salt

c. helium

g. Brass

d. Pacific ocean

h. oxygen

Matter And Energy Study Guide

Roman Wölfel

Matter And Energy Study Guide:

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