$50 + x^2 - \sqrt{5}x + 1 = 0$ ever $x + \frac{1}{x} = 300$?

(₹) - 2√5

(♥) -√5 (♥) 5√5

১৬। x + y = 6 এবং x - y = 4 হলে 5xy এর মান কত?

(*) 16

(M) VS

(3) 24

(41) 25

(N) 36

১৭। U = {1, 2, 3, 4, 5, 6, 7}, A = {3, 5, 6, 7} হলে এং সমান কভ হবেঃ

(*) (1, 2, 4)

(4) (2, 4, 5)

(4) (2, 4, 6)

(N) (3, 5, 6, 7)

56: A = (3, 4, 5) are B = (4, 5, 7) ROW ... L. A \cup B = (3, 4, 5, 7) H. A \cap B = (4, 5)

III. $A \subset B$

নিচের কোনটি সঠিক?

(本) / 电 //

(4) 1 9 111

(M) 11 @ 111 (M) 1, 11 @ 111

১৯। x² – x – 42 এর উৎপানকে বিশ্লেখিক রূপ নিচের কোনটি?

(>) (x - 6)(x - 7)

(4) (x-6)(x+7)

(9) (x+6)(x+7)(8) (x-7)(x+6)

২০। কোনো বিশুটি x-অক্ষের উপর অবস্থিত?

(季) (0, -4)

(⋖) (-4,0)

(4) (0, 3)

(N) (3, -1)

২১। কোন প্রকৃত ভগ্নাংশের লব ও হরের যোগফল 14 এবং বিয়োগফল ৪ হলে, ভগ্নাংশটি নিচের কোনটিঃ

(क) ≥

(₹) ≥ 5

(a) $\frac{11}{2}$

(8) $\frac{11}{3}$

২২ ৷ $m^2 - 6m - 1 = 0$ হলে $m^2 + \frac{1}{m^2} = কজ?$

(m) 32

· (×) 34

(41) 36

(3) 38

২০। একটি সমন্তত্মিক বেলনের ব্যাসার্থ 4 সে.মি. এবং উচ্চতা 6 সে.মি.। বেলনটির বক্ত পৃষ্ঠের ক্ষেত্রফল কতঃ (ম = 3-14)

(ক) 75-36 বর্গ সে.মি.

(খ) 150-72 বর্গ সে.মি.

(গ) 226-08 বর্গ সে.মি.

(খ) 301-44 বর্গ সে.মি.

নিচের চিত্র অনুসারে ২৪ ও ২৫নং প্রশ্নের উত্তর দাও :



ABCD বৰ্গক্ষেত্ৰের কর্মন্তর 'O' বিন্দৃতে মিলিত হয়েছে এবং $BC=6~\mathrm{cm}$.

28 | ZAOB = \$57

(₹) 30°

(31) 45°

(*I) 60°

(N) 90°

२व । AC अब देशकी करा

(ক) 6 লে.মি.

(খ) 6√2 েশ.খি.

(গ) 9√2 সে.মি.
(খ) 12 সে.মি.
২৬। যে চতুর্বজের দুই জোড়া সন্নিহিত বাহ সমান তাকে কী

বলে? (ক) আয়ত

(খ) বৰ্গ**ি**

(গ) ছকি

(খ) সামাজরিক



ABC ত্রিভুজের ক্ষেত্রকল কত?

(ক) 24 বর্গ সে.মি.

(খ) 36 বর্গ সে.মি.

(গ) 48 বর্গ সে.মি.

(খ) 60 বর্গ সে.মি.

২৮। একটি ত্রিভুজার বাহুওলোর অনুপাত $x:x:x\sqrt{2}$ -হলে, এর বৃহত্তম কোশটির মান কতঃ

(*) 80°

(N) 85°

(°1) 90°

(N) 120°

২৯। চতুর্ভুজের চারটি কোণের সমষ্টি কত?

(季) 360°

(N) 300°

(M) 270°

(N) 180°

ত০। একটি ট্রাপিজিয়ামের সমান্তরাল বাহু দুইটির দৈর্ঘ্য ফ্রথাক্রমে 13 সে.মি. এবং ৪ সে.মি। সমান্তরাল বাহুরয়ের লম্ব দুরত্ব 4 সে.মি. হলে ট্রাপিজিয়ামটির ক্ষেত্রফল কত বর্গত

(ক) 42 বর্গ সে.মি.

(খ) 52 বর্গ সে.মি.

(গ) 84 বর্গ সে.মি.

(ঘ) 104 বর্ণ সে.মি.

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web what does the crossed arrow represent in the comic strip 11 what are two of the definitions of dipole given in the comic strip 12 making sense 13 if you finish early what does electronegativity have to do with polarity using polar bears and penguins create an illustration showing a hydrogen sulfide molecule h2s

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web each dipole represents one polar bond the molecule is nonpolar 10 describe what you think is happening to the penguins in the illustration of h 2 o on page 4 they are being pulled towards the stronger polar bear he s more electronegative 11 explain what you think the crossed arrow represents in the comic book polar bonds perhaps 12

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web lesson 3 3 3 polar bears and penguins 43 lesson 3 3 4 thinking electronegatively 50 this unit could also be called organic chemistry which includes studying the kinds of compounds that are essential for living organisms answer the following question cexamine the following molecules what patterns do you see in the bonding of

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web the two polar bears or penguins are equal in electronegativity therefore they pull electrons equally as hard and neither one will end up with more electrons more of the time 4 a what 3 types of bonds are represented on the third page of the comic book nonpolar covalent polar covalent and ionic b

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web lesson 16 polar bear and penguins this shows how the two atoms are bonded together with the two scoop of ice cream representing the bonded pairs of electrons the polar bear and penguin

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web answer because the two polar bears or penguins have equal electronegativity they will pull electrons equally hard and in most cases neither will end up with more electrons each polar bear and penguin has an electronegativity of the same value a non polar bond is formed when two polar bears or two penguins are bonded together

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web why does the polar bear tell the penguin to try to be more positive the polar bear benefits more when the penguin is

more positive the more the polar bear attracts the shared pair of electrons the more positive the partial charge on the penguin

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web to answer this question you will examine 1 electronegativity 2 nonpolar vs polar molecules 3 electronegativity and bonding electronegativity the h atom and the cl atom in hydrogen chloride

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