


## How Substances Dissolve

### KEY IDEAS

As you read this section, keep these questions in mind:


- Why is water called the universal solvent?
- How do substances dissolve?

### Why Do Substances Dissolve in Certain Solvents?

Water is often called a universal solvent because many substances dissolve in it. However, no one substance can dissolve every solute. A general rule in chemistry is that “like dissolves like.” This rule means that a solvent will dissolve substances that have similar molecular structures. 

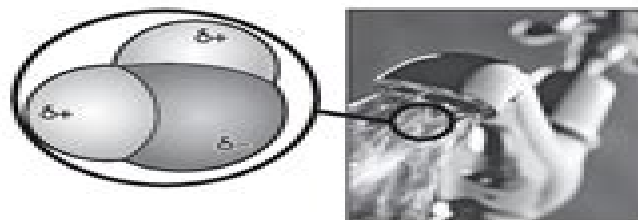
A solvent must be able to attract solute particles and pull them away from one another. Thus, solvent particles must attract solute particles more strongly than the solute particles attract one another.

#### POLAR MOLECULES

Recall that atoms bonded together do not all share electrons equally. For example, a water molecule is made up of two hydrogen atoms bonded to one oxygen atom. The oxygen atom attracts electrons more strongly than the hydrogen atoms do. Thus, the electrons spend more time near the oxygen atom. This gives each hydrogen atom a partial positive charge ( $\delta^+$ ) and the oxygen atom a partial negative charge ( $\delta^-$ ). 

Molecules that have partially charged positive and negative areas are **polar**. Because water molecules are polar, water can dissolve many other polar substances. Water can also dissolve ionic compounds such as table salt.

Water is a polar molecule because the oxygen atom strongly attracts electrons. Because the electrons spend more time near the oxygen atom, the oxygen atom has a partial negative charge.



### READING TOOLBOX

**Summarize** Read each page of the section silently to yourself. With a partner, take turns summarizing each page.

#### READING CHECK

**1. Identify** Why is water called a universal solvent?

because  
many substances dissolve  
in it

#### READING CHECK

**2. Explain** Why does a water molecule have partial charges?

The oxygen atom attracts  
electrons more strongly  
than the hydrogen atoms  
do. Thus, the electrons  
spend more time near the  
oxygen atom.

# How Substances Dissolve Answer Key

**Rowena Douglas**



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connections between environmental science and technology Part II develops the air quality principles basic to an understanding of air quality Part III focuses on water quality and the characteristics of water and water bodies water sciences water pollution and water wastewater treatment Part IV deals with soil science and emphasizes soil as a natural resource highlighting the many interactions between soil and other components of the ecosystem Part V is devoted to showing how decisions regarding handling solid and hazardous waste have or can have profound impact on the environment and the three media discussed in this text air water and soil Finally the epilogue looks at the state of the environment past present and future The emphasis in this brief unit is on mitigating present and future environmental concerns by incorporating technology into the remediation process not by blaming technology for the problem Explaining Primary

Science Paul Chambers,Nicholas Souter,2017-03-25 Successful science teaching in primary schools requires a careful understanding of key scientific knowledge This book covers all the major areas of science relevant for beginning primary school teachers explaining key concepts from the ground up helping trainees develop into confident science educators Classroom activities and Videos of useful science experiments and demonstrations for the primary classroom are integrated into each chapter to translate concepts into teaching practice Chapter content is linked to the National Curriculum in England and the Curriculum for Excellence demonstrating how you could relate understanding to the relevant curriculum taught in schools **Contemporary Advances in Science and Technology Vol-1** Raman Singh,Bhawna

Pareek,2021-09-30 The present volume of Contemporary Advances in Science Technology is focused on advances in chemical and Biological Sciences These includes Pesticides Medicinal Plants Utilized in Marketed anti Arthritic Formulations Inorganic Ion Exchangers Organic Farming Ethical and Practical Implications of Biological Patents Nanomaterials and its Synthesis and Characterization 4 Thiazolidinones Derivatives Impact of COVID 19 Hippuric Acid and Acetohydrazid **Chemistry for**

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DK,2020-06-09 A fantastic aid for coursework homework and studying for tests this comprehensive guide covers Next Generation Science Standards for grades 6 10 and will have you ready for tests and exams in no time Each topic is fully illustrated to support the information make the facts crystal clear and bring the science to life A large central image explains the idea visually and each topic is summed up on a single page helping children to quickly get up to speed and really understand how chemistry works Information boxes explain the theory with the help of simple graphics and for further studying a handy Key Facts box provides a simple summary you can check back on later With clear concise coverage of all the core topics SuperSimple Chemistry is the perfect accessible guide to chemistry for children supporting classwork and

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*Scientifica* David Sang,Peter Ellis,Derek McMonagle,2004 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities      Making Sense of Secondary Science Rosalind Driver,Peter Rushworth,Ann Squires,Valerie Wood-Robinson,2004-03-10 When children begin secondary school they already have knowledge and ideas about many aspects of the natural world from their experiences both in primary classes and outside school This collection of support materials is designed especially for teachers of the early years in secondary school to give guidance both on the ideas which children are likely to bring with them and also on using these ideas to help pupils to make sense of their experiences in science lessons The materials are in 24 sections structured around three themes life and living processes materials and their properties and physical processes Included in each section is a science map identifying key science ideas and also a set of learning guides which give detailed advice on helping children to develop these ideas Written in collaboration with teachers field tested in schools and suitable for use with any published science scheme these materials will be an essential resource for all science teachers who are planning teaching schemes and developing science lessons within the National Curriculum A separate paperback Making Sense of Secondary Science Research into Children s Ideas comes with the file and is also available separately This provides a summary of research in the area and a detailed bibliography for those who want to

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