

(Lessons 10-1 through 10-4)

**Part I** Write the letter for the correct answer in the blank at the right of each question.

1. What is the name of the longest chord in a circle?

A. diameter      B. radius      C. secant      D. tangent

1. \_\_\_\_\_

2. The radius of
- $\odot B$
- is 4 centimeters and the circumference of
- $\odot A$
- is
- $20\pi$
- centimeters. Find
- $CD$
- .

F. 10 cm      H. 24 cm  
G. 14 cm      J. 28 cm

2. \_\_\_\_\_



3. A chord of
- $\odot P$
- measures 8 inches and the distance from the center to the chord is 3 inches. Find the radius of
- $\odot P$
- .

A. 3 in.      B. 5 in.      C.  $\sqrt{73}$  in.      D. 10 in.

3. \_\_\_\_\_

4. If
- $m\angle MON = 86^\circ$
- , find
- $m\angle MPN$
- .

F.  $86^\circ$       H.  $43^\circ$   
G.  $45^\circ$       J.  $30^\circ$ 

4. \_\_\_\_\_



5. Find
- $x$
- if
- $m\angle 1 = 2x + 10$
- and
- $m\angle 2 = 3x - 6$
- .

A. 4      C. 24  
B. 16      D. 42

5. \_\_\_\_\_

**Part II**

- 6.
- $\overline{AE}$
- is a diameter of
- $\odot G$
- and
- $m\angle BGE = 136^\circ$
- . Find
- $m\widehat{AB}$
- .

6. \_\_\_\_\_



7. A circle with a radius of 12 inches has an arc that measures
- $8\pi$
- inches. Find the measure of the central angle determined by this arc.

7. \_\_\_\_\_

8. In
- $\odot P$
- , chord
- $\overline{AB}$
- measures
- $4x - 6$
- centimeters and chord
- $\overline{CD}$
- measures
- $6x - 12$
- centimeters. If
- $\overline{AB}$
- and
- $\overline{CD}$
- are each 4 centimeters from
- $P$
- , find
- $AP$
- .

8. \_\_\_\_\_

9. A 15-inch by 8-inch tablecloth is placed on a circular table. Each of the four corners of the tablecloth touch the edge of the table. Determine the radius of the table.

9. \_\_\_\_\_

10. Quadrilateral
- $ABCD$
- is inscribed in
- $\odot P$
- . Find
- $m\angle ABC$
- .

10. \_\_\_\_\_



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web never true in the quantum mechanical model of the atom the probability of finding an electron within a certain volume of space surrounding the nucleus can be portrayed as a fuzzy cloud always true the fourth principal energy level of an atom donations 32 electrons sometimes true

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web limit an electron s energy to certain values differences between bohr s model and quantum mechanical model of the atom q mechanical model of the atom makes no attempt to describe the electron s path around the nucleus where bohr s model does atomic orbital