## **CLASSIFYING CHEMICAL REACTIONS**

14. SO<sub>3</sub> + H<sub>2</sub>O → H<sub>2</sub>SO<sub>4</sub>



Classify the following reactions as synthesis, decomposition, single replacement or double replacement. (AACCI = RC)

| rep | acement.  | ANSW         | EKS/             |
|-----|---|--------------|------------------|
| 1.  | 2KCIO <sub>3</sub> → 2KCI + 3O <sub>2</sub>                             | Decon        | nposition        |
| 2.  | HCI + NaOH → NaCI + H₂O   | Double       | Replacement      |
| з.  | Mg + 2HCl → MgCl <sub>2</sub> + H <sub>2</sub>                          | Single       | Replacement.     |
| 4.  | 2H <sub>2</sub> + O <sub>2</sub> → 2H <sub>2</sub> O                    | Synthesis    |                  |
| 5.  | 2AI + 3NIBr₂ → 2AIBr₃ + 3NI   | Single       | Replacement      |
| 6.  | 4AI + 3O₂ → 2AI₂O₃  | Synth        | re Sis           |
| 7.  | 2NaCl → 2Na + Cl <sub>2</sub>   | Decom        | position         |
| 8.  | $CaCl_2 + F_2 \rightarrow CaF_2 + Cl_2$                                 | Single       | Replacement      |
| 9.  | AgNO <sub>3</sub> + KCI → AgCI + KNO <sub>3</sub>                       | Double       | Replacement      |
| 10. | N <sub>2</sub> + 3H <sub>2</sub> → 2NH <sub>3</sub>                     | Synth        | nesis            |
| 11. | 2H <sub>2</sub> O <sub>2</sub> → 2H <sub>2</sub> O + O <sub>2</sub>     | Decomp       | position         |
| 12. | $(NH_4)_2SO_4 + Ba(NO_3)_2 \rightarrow BaSO_4$                          | + 2NH4NO3 DO | uble Replacement |
| 13. | Mgl <sub>2</sub> + Br <sub>2</sub> → MgBr <sub>2</sub> + l <sub>2</sub> | Single       | Replacement      |

Synthesis

# **General Chemistry Massing Activities Lab Answers**

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