

CASCADED MC QUESTIONS: GENERAL CHEMISTRY II

100 Pts (200 Pts Score) (20 Pts per Question)

1. Which among the intermolecular forces is the weakest?
 - A. Hydrogen Bond
 - B. Ion-Dipole Forces
 - C. Dipole-Dipole Forces
 - D. Dispersion Forces
2. In which of the following would dipole-dipole forces exist?
 - A. Butane
 - B. Methanol
 - C. CH_2Cl_2
 - D. CH_3I
3. The vapor pressure of ethanol at 40 °C is 0.123 kPa. What would be the vapor pressure at 30 °C?
 - A. 0.123 kPa
 - B. Greater than 0.123 kPa
 - C. Less than 0.123 kPa
 - D. Cannot be determined
4. Which of the following statements about the following process will *not* result in a higher yield?
 - A. All the products of the reaction
 - B. Increased
 - C. It should take some time to reach equilibrium of reaction
 - D. It depends on the nature of the
5. Which of the following would have the smallest ΔH_{vap} ?
 - A. H_2O
 - B. CH_3OH
 - C. H_2
 - D. CH_3Cl
6. Which of the following reactions would be affected by the addition of H_2O ?
 - A. $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
 - B. $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
 - C. $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
 - D. $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
7. Which of the following describes Henry's Law?
 - A. Partial pressure increases with the solubility of gases.
 - B. Partial pressure increases with the solubility of gases.
 - C. Partial pressure increases with the solubility of gases.
 - D. Partial pressure increases with the solubility of gases.
8. How much of the product is formed at 25 °C if 1.00 mol of H_2O is added to 1.00 mol of H_2O ?
 - A. 1.00 mol
 - B. 1.00 g
 - C. 1.00 mol
 - D. 1.00 g
9. Water's vapor pressure at 25 °C is 0.031 atm. The temperature is 25 °C. The vapor pressure is 0.031 atm. The temperature is 25 °C. The vapor pressure is 0.031 atm.
 - A. 0.031 atm
 - B. 0.031 atm
 - C. 0.031 atm
 - D. 0.031 atm
10. Which of the following is the correct statement?
 - A. The vapor pressure of water is 0.031 atm at 25 °C.
 - B. The vapor pressure of water is 0.031 atm at 25 °C.
 - C. The vapor pressure of water is 0.031 atm at 25 °C.
 - D. The vapor pressure of water is 0.031 atm at 25 °C.

General Chemistry Diagnostic Test

Marco Casella



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