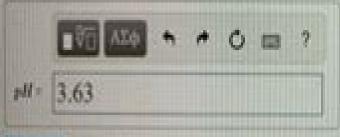


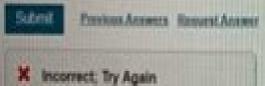
Acid Base Homework

Exercise 16.69

## · Part A

If 20.0 mL of glacial acetic acid (pure  $HC_1H_2O_2$ ) is distend to 1.50 L with water, what is the pH of the resulting solution? The density of glacial acetic acid is 1.05 g/mL. Express your answer to two decimal places.







# **General Acid And Base Homework Answers**

Marcel A. Müller

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