

## Question 4

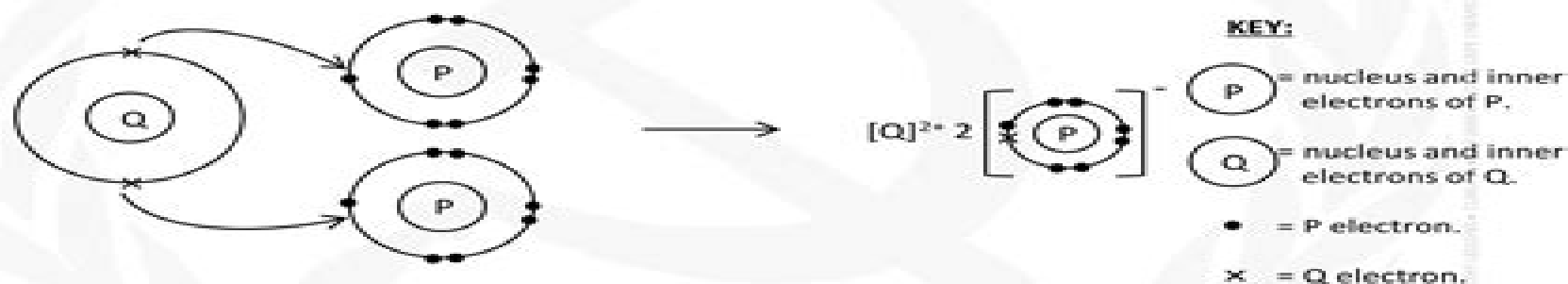
- 4(a) Radiocarbon dating using  $^{14}\text{C}$  is a method of determining the age of once living, archaeological specimens which are no more than 50 000 years old.

$^{127}\text{I}$  is an example of a substance which can be used as a tracer. Tracers which concentrate in some organs are introduced into the bodies of living organisms and used to study chemical changes.

- 4(b)(i) Number of occupied electron shells = period number.  
Number of electrons in outer shell = group number.

	P	Q	R	S
Group #	7	2	6	2
Period #	2	3	3	4

- 4(b)(ii) The compound is ionic.



Formula is  $\text{QP}_2$ .

- 4(b)(iii) S will react more vigorously with hydrochloric acid than Q because S is lower in group 2 so it loses its outer electrons more easily than Q.

Equation for reaction of Q with HCl:  $\text{Q}_{(s)} + 2\text{HCl}_{(aq)} \rightarrow \text{QCl}_{2(aq)} + \text{H}_{2(g)}$

# F32chemistry June 2014 Paper

**R Barnett**



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## **Table of Contents F32chemistry June 2014 Paper**

1. Understanding the eBook F32chemistry June 2014 Paper
  - The Rise of Digital Reading F32chemistry June 2014 Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying F32chemistry June 2014 Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an F32chemistry June 2014 Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from F32chemistry June 2014 Paper
  - Personalized Recommendations

- F32chemistry June 2014 Paper User Reviews and Ratings
- F32chemistry June 2014 Paper and Bestseller Lists
- 5. Accessing F32chemistry June 2014 Paper Free and Paid eBooks
  - F32chemistry June 2014 Paper Public Domain eBooks
  - F32chemistry June 2014 Paper eBook Subscription Services
  - F32chemistry June 2014 Paper Budget-Friendly Options
- 6. Navigating F32chemistry June 2014 Paper eBook Formats
  - ePub, PDF, MOBI, and More
  - F32chemistry June 2014 Paper Compatibility with Devices
  - F32chemistry June 2014 Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of F32chemistry June 2014 Paper
  - Highlighting and Note-Taking F32chemistry June 2014 Paper
  - Interactive Elements F32chemistry June 2014 Paper
- 8. Staying Engaged with F32chemistry June 2014 Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers F32chemistry June 2014 Paper
- 9. Balancing eBooks and Physical Books F32chemistry June 2014 Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection F32chemistry June 2014 Paper
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine F32chemistry June 2014 Paper
  - Setting Reading Goals F32chemistry June 2014 Paper
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of F32chemistry June 2014 Paper
  - Fact-Checking eBook Content of F32chemistry June 2014 Paper

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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