

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ relationship _____ equated _____

2. $40g = 40r + 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ from the _____ involved in the reaction _____

3. Problem

When aluminum chloride reacts with aluminum hydroxide, aluminum chloride is produced



moles of

$2AlCl_3$	$3H_2O$	$2Al(OH)_3$	$3HCl$
$2AlCl_3$	$3H_2O$	$2Al(OH)_3$	$3HCl$
$2AlCl_3$	$3H_2O$	$2Al(OH)_3$	$3HCl$

2. How many moles of $AlCl_3$ will be produced?

2. $2AlCl_3 + 3H_2O \rightarrow 2Al(OH)_3 + 3HCl$

Episode 801 Chemistry Answer Key

Kathleen Armour

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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