## **ORGANIC CHEMISTRY**

#### KIPS

The group which activates the benze	ne ring except
aOH	b OCH <sub>3</sub>
c NH <sub>2</sub>	$\mathbf{d}_{i} - \mathbf{C} = \mathbf{N}$
Which may not present as beteroato	m in heterocyclic compounds generally
a. Sulphur	b. Oxygra
c. Nitrogen	d. Hromane
The IUPAC name of the following co	empound is
H <sub>1</sub> CCH <sub>2</sub> (CH <sub>3</sub> )	C(CaHa)GHaCHaCHa
a. 3-Ethyl-3-methylbexane	b. 2-Ethyl-3-methylhexane
c. 3-Methyl-3-ethylhexane	d. 4 Ethyl-2-methylhexane
The nucleophilicity of benzene ring i	s increased if group is attached to it
aCOR	bCHi
cCN	dCI
The common name of the compound	CH2CCOOH): is
a. Malonic acid	b. Propan -1, 3-dioic acid
c. Propen -1, 3-oie acid	d. Both "A" and "C"
is less reactive than benzeue	although it contains or the, para directing group
a. Nitrobenzene	b. Phenol
e. Chlorobentene	d. Benzene sulphonic acid
The bond between carbon 2 and 3 in	the CH - CH = CH2 involves the hybridization
a, sp² and sp²	b, sp and sp
c. sp and sp <sup>3</sup>	id. isp <sup>2</sup> and sp <sup>3</sup>
The most stable carbanion is	
a. CH;	b. (CH <sub>2</sub> ) <sub>2</sub> CH <sup>-</sup>
c. (CH <sub>2</sub> ) <sub>2</sub> C	d. CH <sub>2</sub> CH <sub>2</sub>
Which structures show a primary al	cohol that cannot be dehydrated to form an alkene
1 CH-OH	
II CH-CH-OH	
	b. Only II and III
	d. Only I and III
	b. o-mitrotolmene
	d. 2,4,6-trinitrotoluene
	b. 6
	d. 8
	<ul> <li>b. Have unshared electrons pairs</li> </ul>
	d. Are negatively charged
Which alkene could exist in civ and t	The section of the se
a. CH, CH, CH, CH = CH,	CH,CH,C - CH,
	c. — NH <sub>2</sub> Which may not present as beteroato a Sulphur c. Nitrogen  The IUPAC name of the following co H <sub>3</sub> CCH <sub>2</sub> (CH <sub>3</sub> )  a. 3—Ethyl—3—methylbecame c. 3—Methyl—3—ethylbecame  The nucleophilicity of benzene ring is a.—COR  c.—CN  The common name of the compound a Malonic neid c. Propan—1, 3—oie seid  is less reactive than benzene a Nitrobenzene  c. Chlorobenzene  The bond between garbon 2 and 3 in a sp² and sp²  c. sp and sp²  c. sp and sp²  The most stable carbanion is a CH <sub>3</sub> c. (CH <sub>3</sub> ) <sub>3</sub> C <sup>-</sup> Which structures show a primary all CH <sub>3</sub> OH

# **Environntal Chemistry Mcq**

**S Ben Porath** 

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