

Assignment #1 – Compound Names and Formulas
Single-valent ions only

A. Name these Compounds

1. Li_2S Lithium Sulfide
2. CaO Calcium Oxide
3. NaF Sodium Fluoride
4. CaBr_2 Calcium Bromide
5. MgCl_2 Magnesium Chloride
6. BBr_3 Boron Bromide
7. Cs_2O Cesium Oxide
8. FrBr Francium Bromide
9. Ag_2S Silver Sulfide

10. GeF_4 Germanium Fluoride
11. Ga_2O_3 Gallium Oxide
12. EsCl_3 Einsteinium Chloride
13. Fm_2O_3 Fermium Oxide
14. Mg_3N_2 Magnesium Nitride
15. Rb_2O Rubidium Oxide
16. RaO Radium Oxide
17. SrO Strontium Oxide
18. Tc_2O_7 Technetium Oxide

B. Write the correct chemical formula for these compounds by balancing the ionic charges

1. sodium chloride NaCl
2. magnesium fluoride MgF_2
3. silver oxide Ag_2O
4. indium bromide InBr_3
5. zinc bromide ZnBr_2
6. neodymium oxide Nd_2O_3
7. thorium sulphide ThS_2
8. actinium oxide Ac_2O_3
9. radium bromide RaBr_2
10. cesium oxide Cs_2O

11. hydrogen oxide H_2O
12. francium nitride Fr_3N
13. rubidium phosphide Rb_3P
14. potassium oxide K_2O
15. beryllium sulphide BeS
16. lithium sulphide Li_2S
17. hydrogen bromide HBr
18. strontium nitride Sr_3N_2
19. calcium oxide CaO
20. tantalum nitride Ta_3N_5

Naming Compounds And Writing Formulas Answer Key

**Richard Post, Chad Snyder, Clifford C.
Houk**



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Stoichiometry Unit Project Luann Marie Decker, 1998

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Table of Contents Naming Compounds And Writing Formulas Answer Key

1. Understanding the eBook Naming Compounds And Writing Formulas Answer Key
 - The Rise of Digital Reading Naming Compounds And Writing Formulas Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Naming Compounds And Writing Formulas Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming Compounds And Writing Formulas Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Compounds And Writing Formulas Answer Key
 - Personalized Recommendations
 - Naming Compounds And Writing Formulas Answer Key User Reviews and Ratings
 - Naming Compounds And Writing Formulas Answer Key and Bestseller Lists
5. Accessing Naming Compounds And Writing Formulas Answer Key Free and Paid eBooks
 - Naming Compounds And Writing Formulas Answer Key Public Domain eBooks
 - Naming Compounds And Writing Formulas Answer Key eBook Subscription Services
 - Naming Compounds And Writing Formulas Answer Key Budget-Friendly Options
6. Navigating Naming Compounds And Writing Formulas Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Naming Compounds And Writing Formulas Answer Key Compatibility with Devices
 - Naming Compounds And Writing Formulas Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming Compounds And Writing Formulas Answer Key
 - Highlighting and Note-Taking Naming Compounds And Writing Formulas Answer Key
 - Interactive Elements Naming Compounds And Writing Formulas Answer Key
8. Staying Engaged with Naming Compounds And Writing Formulas Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Naming Compounds And Writing Formulas Answer Key
- 9. Balancing eBooks and Physical Books Naming Compounds And Writing Formulas Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Naming Compounds And Writing Formulas Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Naming Compounds And Writing Formulas Answer Key
 - Setting Reading Goals Naming Compounds And Writing Formulas Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Naming Compounds And Writing Formulas Answer Key
 - Fact-Checking eBook Content of Naming Compounds And Writing Formulas Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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