| Name: | KEY | Block: |
|--|-----|--------|
| The state of the s | | |

11. MOBr. Malybadenium (II) Bramide

12. VCI Vanadium (V) Chlaride

13. Mn2O3 Manganese (III) Oxide

15. NP2O3 NEDTUNIUM (II) OXICE

16. V2Os Vanadium (V) Oxide

14. COO _ COLONH (II) Oxide

Assignment #2 - Compound Names and Formulas Multi-valent ions only

| 1. copper (II) oxide | 11. manganese (III) oxide Mr 203 |
|--|--|
| 2. mercury (I) oxide +\0,20 | 12. vanadium (II) bromide VB(z |
| B. gold (III) chloride _AuCl3 | 13. niobium(V) oxide |
| 4. thallium (III) bromide TIPY 3 | 14. titanium (III) oxide T\20> |
| 5. bismuth (V) oxide <u>B1205</u> | 15. titanium (III) nitride TIN |
| 3. terbium (III) oxide | 16. iron (II) oxide FeO |
| . uranium (VI) oxide \(\lambda \) | 17. cobalt (II) phosphide C03P2 |
| protactinium (IV) bromide PhBY4 | 18. tin (II) oxide |
| cerium (III) oxide | 19. thulium (II) bromide _\TWBY ≈ |
| 10. arsenic (V) sulphideA52S5 | 20. copper (II) bromide CUBY2 |
| Name the compound using the Roman Nume | eral Method and balance the ionic charges. |
| . snc4 Tin (IV) Chloride | 9. PdF. Palladium (IV) Fluoride |
| 2. BIBIS _ BISMINTO (V) Bromide | 10. Osoo Osmium (III) Oxide |

3. POO2 POLONIUM (IV) OXICE

6. HgCI <u>HEYCUYY</u> (I) Chloride

5. Hgo Mercury (II) Oxide

4. Pblz _ I cod(I) Iodide

7. Auzos Gold (III) Oxide.

8. FeCI Iron (II) Chlorida

Names And Formulas Worksheet 1

Paul McFedries, Greg Harvey

Names And Formulas Worksheet 1:

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