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Assignment #2 – Compound Names and Formulas Multi-valent ions only

A. Write the correct formula for the	e following compounds, all of which have been named using the
	The ionic charge is given after the first element (metallic) in Roman
0 0	

1. copper (II) oxide	11. manganese (III) oxide MY 203
2. mercury (I) oxide	12. vanadium (II) bromide VBC
3. gold (III) chloride AuCl3	13. niobium(V) oxide Nb205
4. thallium (III) bromide TIPY 3	14. titanium (III) oxide
5. bismuth (V) oxide <u>B1205</u>	15. titanium (III) nitride TIN
6. terbium (III) oxideT_>03	16. iron (II) oxide <u>Fe.O</u>
7. uranium (VI) oxide \(\lambda \)	17. cobalt (II) phosphide C03P2
8. protactinium (IV) bromide PhBY4	18. tin (II) oxide
9. cerium (III) oxide	19. thulium (II) bromide TWBY 2
10. arsenic (V) sulphideAS2S5	20. copper (II) bromide CUBY2
B. Name the compound using the Roman Numer 1. SnCl ₄	9. PdF4 RONOGIAM (IV) FLUOVIDE
2. BIBr. BISMINTH (V) Bromide	10. Os2O3 OSMIUM (III) Oxide
3. POO2 POLONIUM (IV) OXICE	11. MOBr. Malybodenum (II) Bramide
4. Pbl2 100d(II) Iodide	12. VCIs Vanadium (V) Chloride
5. HgO Mercury (II) Oxide	13. Mn2O3 Managamese (III) Oxide
6. Hgci <u>Meycuyy</u> (I) Chloride	14. COO COLONH (II) Oxide
7. AuzOs Gold (III) Oxide	15. NP2O3 NEDTUNIUM (II) OXIC
8. FeCI2 Iron (II) Chicride	16. V2Os Vanadium (V) Oxide

Molecular Geometry Worksheet Answers Pogil

CO Houle

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