#### Honors Chemistry Chapter 7 Exam Review Sheet

- 1. Make sure you have read the entire chapter.
- 2. Be able to define all of the chapter 7 vocabulary words.
  - a. Including:
  - electron dot.
  - structure
  - lonic compound
  - valence electron
  - ionic bond
  - chemical formula
  - halide ion.

- afflow
- octet rule
- formula unit;
- coordination number
- metallic bond
- Know the number of valence electrons (electrons in the highest occupied energy level of an atom) for the elements in the representative groups 1A-7A.
- Know the charges of the ions for the elements in the representative groups 1A-7A.
- The maximum charge an ion is likely to have is -3 or +3.
- Be able to write the electron configurations for both the ions and the atoms in the representative groups 1A-7A and for the transition metals.
  - a. Note: the ions for these elements will have a pseudo noble gas configuration.
- 7. Be able to identify the number of electrons an element gives up or takes on when it forms an ion.
  - Metals donate electrons when they form ions. They will have positive charges. (Cations)
  - Nonmetals take on electrons when they form ions. They will have negative charges. (Anions).
- 8. A pseudo-noble gas for a transition metal contains a full 3d sublevel.
- a. Cu\*, Zn²\*, have a pseudo-noble gas configuration.
- 9. Be able to write the correct formula unit for an ionic compound.
  - The overall charge of a correct formula unit is zero.
  - b. use the criss-cross method to write formulas.
- Be able to write the correct electron configuration for the all of the elements, all of the ions that the elements form and be able to
  identify the correct charge for all of the ions that they form.
  - a. Including but not limited to:
    - Li, Mg, He, Sn, Si, Ca, Ga, Sr, Ag, Ba, Cu, Fe, Ni, K., Mn, Br, L, Cd, N. P. F. O. Cr, Sn<sup>+</sup>, F. Sn<sup>+</sup>, O<sup>2</sup>
- The formula for sodium sulfate is Na<sub>2</sub>SO<sub>4</sub>.
- 12. The condensed (abbreviated) electron configuration distinguishes the core electrons from the valence electrons. CORE electrons are tightly held to the nucleus and resemble a noble gas configuration. VALENCE electrons are the outer most electrons and are involved in chemical reactions.
- 13. Examples of full & condensed (abbreviated) electron configuration:

Full electron configuration		Condensed (abbreviated) electron configuration
86.	Li: 1s2 2s2	Lit[He] 2s <sup>1</sup>
b.	Co: 1s2 2s2 2p4 3s2 3p4 4s2 3d2	Co:[Ar] 4s <sup>3</sup> 3d <sup>3</sup>
40.	Br: 1s2 2s2 2p4 3s2 3p4 4s2 3d24 4p5	Br:[Ar] 4st 3d** 4p*
d.	52: 1s2 2s2 2p4 3s2 3p4	52: [Ne] 3s2 3p4
40	Fe <sup>3+</sup> : 1s <sup>2</sup> 2s <sup>2</sup> 2o <sup>4</sup> 3s <sup>2</sup> 3p <sup>4</sup> 4s <sup>6</sup> 3d <sup>4</sup>	Fe <sup>2+</sup> :[Ar] 4s <sup>0</sup> 3d <sup>0</sup>

- Alloys a mixture composed of two or more elements, at least one of which is a metal. Their properties are often superior to their component elements. EX:
  - a. Bronze is tougher than pure copper
  - Steel resists corrosion & is strong than iron. Steels have a wide range of uses.
  - Some are substitutional alloys and others are interstitial alloys.
- 15. Be able to identify the properties of ionic compounds. Ex:
  - i. They are made of cations & anions bonded together.
  - iii. They are crystalline solids at room temperature
  - iii. They have high melting points
  - ly. They conduct electricity when they are dissolved in water or melted.
- Be able to identify the shape & coordination number of ions in a crystal. The coordination number of an ion is the number of ions
  of opposite charge that surround the ion in a crystal. Ex:
  - a. Body-centered cubic, 8
  - b. Face-centered cubic, 12
  - c. Hexagonal close-packing, 12

# **Modern Chemistry Answers Ch 7 Mixed Review**

David L. Bish, Jeffrey E. Post

# Modern Chemistry Answers Ch 7 Mixed Review:

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Current Trends and Future Developments on (Bio-) Membranes Angelo Basile, Evangelos P. Favvas, 2023-11-21 Modern Approaches in Membrane Technology for Gas Separation and Water Treatment presents condensed information on novel and promising membrane materials. The book answers some major questions from the membrane community about three promising materials that are to be introduced at industrial scale. It introduces recent out of the box ideas concerning the application of new methods capable to enhance the membrane separation efficiency Sections cover potential commercialization important question on three famous membrane materials and new approaches in membrane technology Finally the book describes and discusses three novel ideas about the potential effect of the magnetic field on membrane separation efficiency the use of cryogenic technology on membrane separations and the use of nanobubble technology on water membrane processes Focuses on the necessity for environmental friendly and cost effective purification and separation process Lists all new membrane materials suitable for commercialization Presents new modern approaches and ideas for improving the membrane efficiency

Introduction to Industrial Polypropylene Dennis B. Malpass, Elliot Band, 2012-07-02 This introductory text is an important resource for new engineers chemists students and chemical industry personnel to understand the technical aspects of polypropylene which is the 2nd largest synthetics polymer in manufactured output The book considers the following topics What are the principal types of polypropylene and how do they differ What catalysts are used to produce polypropylene and how do they function What is the role of cocatalysts and how have they evolved over the years How are industrial polypropylene catalysts tested and the resultant polymer evaluated What processes are used in the manufacture of polypropylene What are the biopolymer alternatives to polypropylene What companies are the major industrial manufacturers of polypropylene What is the environmental fate of polypropylene 19th Natural Philosophy Alliance Proceedings Greg Volk, 2012-07-14 The Natural Philosophy Alliance NPA sponsors regular international conferences for presenting high quality papers discussing aspects of philosophy in the sciences Many papers offer challenges to accepted orthodoxy in the sciences especially in physics Everything from the micro physics of quantum mechanics to the macro physics of cosmology is entertained Though the main interest of the NPA is in challenging orthodoxy in the sciences it will also feature papers defending such orthodoxy Our ultimate propose is to enable participants to articulate their own understanding of the truth All papers are reviewed by society officers and sometimes by other members before presentation in conferences and they are edit sometimes very significantly prior to publication in the Proceedings of the NPA

**Fundamentals of Modern Manufacturing** Mikell P. Groover,2010-01-07 Engineers rely on Groover because of the book s quantitative and engineering oriented approach that provides more equations and numerical problem exercises The fourth edition introduces more modern topics including new materials processes and systems End of chapter problems are

also thoroughly revised to make the material more relevant Several figures have been enhanced to significantly improve the quality of artwork All of these changes will help engineers better understand the topic and how to apply it in the field

Modern Powder Diffraction David L. Bish, Jeffrey E. Post, 2018-12-17 Volume 20 of Reviews in Mineralogy attempted to 1 provide examples illustrating the state of the art in powder diffraction with emphasis on applications to geological materials 2 describe how to obtain high quality powder diffraction data and 3 show how to extract maximum information from available data In particular the nonambient experiments are examples of some of the new and exciting areas of study using powder diffraction and the interested reader is directed to the rapidly growing number of published papers on these subjects Powder diffraction has evolved to a point where considerable information can be obtained from ug sized samples where detection limits are in the hundreds of ppm range and where useful data can be obtained in milliseconds to microseconds We hope that the information in this volume will increase the reader's access to the considerable amount of information contained in typical diffraction data Introduction to Biofuels David M. Mousdale, 2010-07-29 What role will biofuels play in the scientific portfolio that might bring energy independence and security revitalize rural infrastructures and wean us off of our addiction to oil The shifting energy landscape of the 21st century with its increased demand for renewable energy technology poses a worrying challenge Discussing the multidisciplin Lab Experiments Modern Chemistry Harold Clark **Independent and Weekly Review**,1917 Metcalfe, 1982 India Rubber World and Electrical Trades Review .1927

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