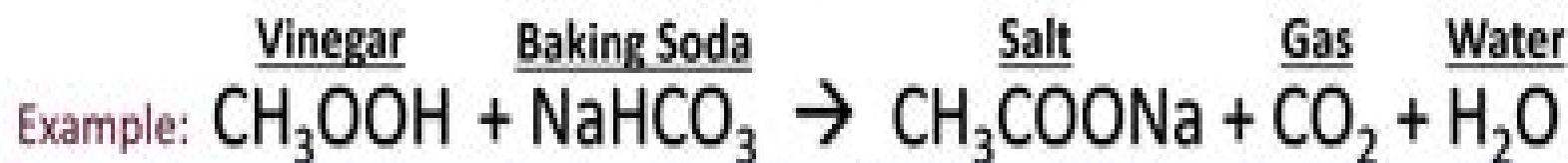
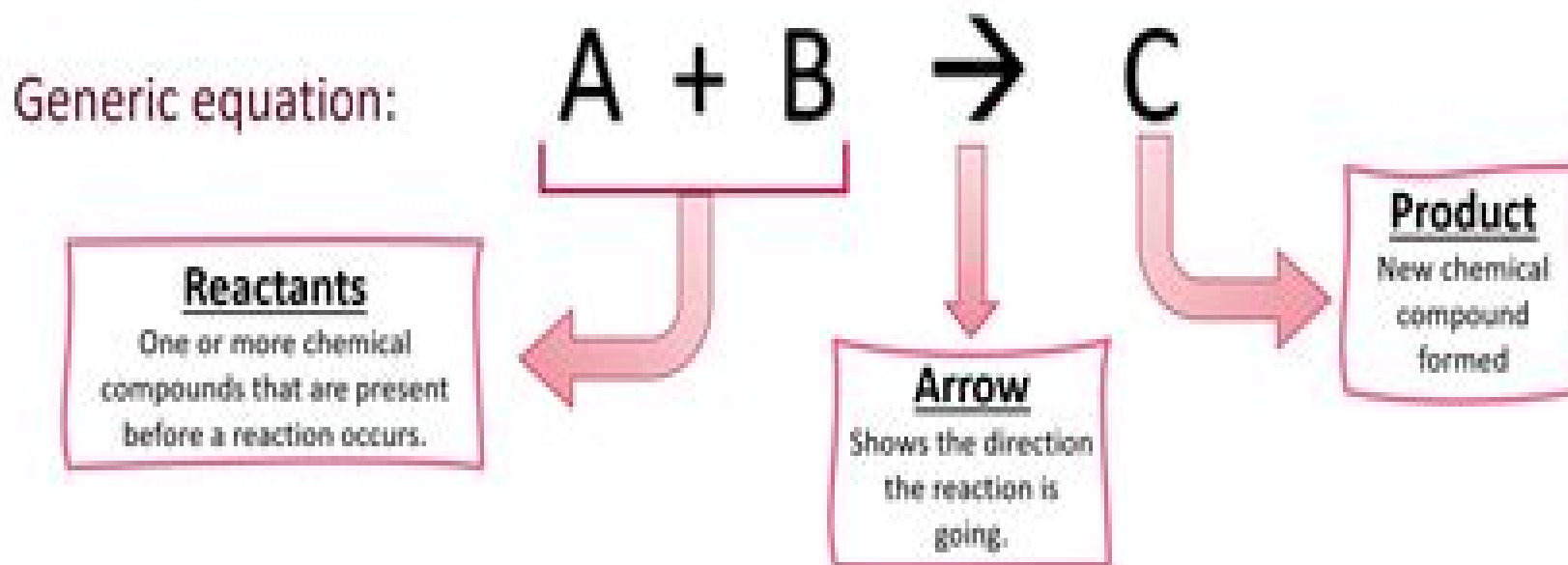


General Chemical Reaction Equations

Chemical reactions are often represented by equations. These equations give information about the reaction in a simplified way.



An example reaction is vinegar/acetic acid reacting with baking soda/sodium bicarbonate to form a salt/sodium acetate, gas/carbon dioxide, and water. This reaction can safely be done at home using ingredients found in your kitchen!

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were recorded and written up by Dr Klaus Muller as a 293 page manuscript At a later stage A K N R joined the effort it was decided to make a fresh start and to write a much more comprehensive text Of methods for direct energy conversion the electrochemical one is the most advanced and seems the most likely to become of considerable practical importance Thus conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met Cor sion is recognized as having an electrochemical basis The synthesis of nylon now contains an important electrochemical stage Some central biological mechanisms have been shown to take place by means of electrochemical reactions A number of American organizations have recently recommended greatly increased activity in training and research in electrochemistry at universities in the United States

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