Section A

For each question there are four possible answers A, B, C and D. Choose the one you consider to be correct.

Use of the Data Booklet may be appropriate for some questions.

1 Compound X is an organic compound that contains 30.6% carbon, 3.8% hydrogen, 20.4% oxygen and 45.2% chlorine by mass.

What is the empirical formula of X?

A C₂H₃OCI

B C₂H₄OCI

C C₂H₄OC/

D C₄H₃O₂CI₂

2 A sample of propane, C₃H₈, with a mass of 9.61 g is completely combusted in an excess of oxygen under room conditions.

Which volume of carbon dioxide gas is produced?

A 4,89 dm³

B 5.24 dm³

C 14.7 dm³

D 15.7 dm³

3 Which atom has the same number of electrons as an ammonium ion?

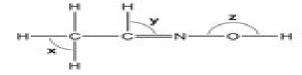
A Mg

B Na

C Ne

D O

4 Ethanal reacts with hydroxylamine, NH₂OH, to form the molecule shown.



What is the order of increasing bond angle in this structure from smallest to largest?

A z. x. y

B y. z. x

C x, z, y

D z, y, x

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