

Isotopes Practice Set

- What is an isotope?
- What does the number next to isotopes signify?
- How can you tell isotopes apart?

For each of the following isotopes, write the number of protons, neutrons, and electrons.

	Chlorine-35	Chlorine-37
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

	Nitrogen-14	Nitrogen-15
# of protons	7	7
# of neutrons	7	8
# of electrons	7	7

	Carbon-12	Carbon-13
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Oxygen-16	Oxygen-18
# of protons	8	8
# of electrons	8	8
# of neutrons	8	10

Fill in the isotopes names including chemical symbol, atomic number, and mass number from the values that your partner gave and the information provided.

	Isotope: 34	Isotope: 35
# of protons	17	17
# of neutrons	17	18
# of electrons	17	17

	Carbon-12	Carbon-13
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Iron-56	Iron-57
# of protons	26	26
# of neutrons	30	31
# of electrons	26	26

	Chlorine-35	Chlorine-37
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

Isotope Practice Set Answer Key

Dr. Ranjit Kumar Singh, IAS (AIR-49)

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

Isotope Practice Set Answer Key:

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5. In chapter 4, the author will scrutinize the relevance of Isotope Practice Set Answer Key in specific contexts. This chapter will explore how Isotope Practice Set Answer Key is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Isotope Practice Set Answer Key. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Isotope Practice Set Answer Key.

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