

CHAPTER
8
Section Quiz
Lessons 8-1 Through 8-3

Choose the best answer.

1. What is the length of the altitude to the hypotenuse in a right triangle if this altitude divides the hypotenuse into segments of lengths 8 millimeters and 18 millimeters?

A 12 mm C 26 mm
 B 13 mm D 36 mm

Use the figure for Exercises 2 and 3.



2. If $XY = 10$ and $XW = 12$, what is VY ?

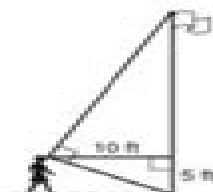
F 2 H 12
 G 4.4 J 14.4

3. If $XY = 9$ and $VW = 6$, what is VX ?

A 3 C 12
 B 4 D 13

4. To estimate the height of a flagpole, Marci, who is 5 feet tall, stands so that her lines of sight to the top and bottom of the pole form a 90° angle. What is the height of the pole to the nearest foot?

F 9 ft H 25 ft
 G 20 ft J 50 ft



5. One acute angle of a right triangle measures 28° . To the nearest tenth, what is the length of the side opposite that angle if the hypotenuse measures 16 meters?

A 7.5 m C 18.1 m
 B 14.1 m D 34.1 m

Use the figure for Exercises 6 and 7.



6. What is $\tan K$?

F $\frac{8}{17}$ H $\frac{15}{17}$
 G $\frac{8}{15}$ J $\frac{15}{8}$

7. To the nearest degree, what is $m\angle M$?

A 2° C 62°
 B 28° D 88°

8. A 24-foot ladder forms a 76° angle with the ground. The top of the ladder rests against a building. To the nearest inch, how high up the building does the ladder reach?

F 5 ft 10 in. H 18 ft 5 in.
 G 18 ft 1 in. J 23 ft 3 in.

9. What is $\cos 30^\circ$?

A $\frac{1}{\sqrt{3}}$ C $\frac{2}{\sqrt{3}}$
 B $\frac{\sqrt{3}}{2}$ D $\sqrt{3}$

10. To the nearest degree, what is the measure of the smallest angle of a right triangle whose legs measure 9 and 15?

F 31° H 53°
 G 37° J 59°

11. To the nearest degree, what angle does a hill with a grade of 11% make with a horizontal line?

A 6° C 79°
 B 11° D 84°

Geometry Chapter 5 Test Form B Answers

James Keeler, Peter Wothers



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Chemical Structure and Reactivity James Keeler, Peter Wothers, 2013-11 Why do certain substances react together in the way that they do What determines the shape of molecules And how can we predict whether a particular reaction will happen at all Such questions lie at the heart of chemistry the science of understanding the composition of substances their reactions and properties Though introductory chemistry is often broken into three sections inorganic organic and physical the only way for students to fully understand the subject is to see it as a single unified whole Chemical Structure and Reactivity rises to the challenge of depicting the reality of chemistry Offering a fresh approach to the subject by depicting it as a seamless discipline the text shows how organic inorganic and physical concepts can be blended together in order to achieve the common goal of understanding chemical systems With a lively and engaging writing style enhanced by vivid illustrations only Chemical Structure and Reactivity makes teaching chemistry with an integrated approach possible Special Features The only introductory text to take a truly integrated approach in explaining the fundamentals of chemistry Fosters an orbital based understanding of reactions with clear curly arrow mechanistic detail throughout A two part structure allows flexibility of use Part I lays down the core of the subject while Part II describes a series of relatively standalone topics which can be selected to fit a particular course Numerous concepts are illustrated with fully cross referenced custom developed online modules enabling students to develop an understanding through active learning Self test exercises embedded in the text with solutions at the end of each chapter and extensive question sets encourage hands on learning to help students master the subject and gain confidence The Online

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