General Chemistry placement test Study Guide 2024 Solved 100%

Atomic Number - The number of protons in the nucleus of an atom. The atomic number of an element never changes, meaning that the number of protons in the nucleus of every atom in an element is always the same.

Hydrogen = 1, Carbon = 6, Oxygen = 8 (located on top of element of periodic table)

Mass Number - # of neurons + # of protons = mass number (located on bottom of periodic table)

 Let's say an atom is missing a neutron or has an extra neutron. That type of atom is called an isotope. An atom is still the same element if it is missing an electron.

Example: (carbon-14)

Distribution 8/W subatomic particles in an Atom - A typical atom consists of three subatomic particles: Protons, Neutrons, and Electrons.

 Nucleons include protons and neutrons. All the positive charge of an atom is contained in the nucleus, and originates from the protons. Neutrons are neutrally-charged. Electrons, which are negatively-charged, are located outside of the nucleus.

Roles of subatomic particles: Protons, Neutrons, & Electrons - Particle Mass:

Protons & Neutrons have about same mass. Electrons have much less mass than protons and neutrons.

Particle Charge:

Protons have a positive charge (+)

- Neutrons have no charge (0)
- Net charge of the nucleus is positive and equal to the number of protons.

General Chemistry Placement Exam Study Guide

Martha Mackin

General Chemistry Placement Exam Study Guide:

Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students MCAT Preparation Guide Miriam S. Willey, Barbara M. Jarecky, 2015-01-13 Competition for admission to American medical schools has always been intense Now with more than 40 000 pre med students applying for the few available slots each year scoring well on the Medical College Admission Test MCAT is more critical than ever The MCAT Preparation Guide offers students a systematic sensible way to improve MCAT test scores It recognizes their need to understand not only the subject areas covered but also the way the MCAT is structured and what test scorers look for Extensively field tested at the University of Chicago Tulane University Howard University and the University of Kentucky this Guide has already helped hundreds of students to boost their MCAT scores significantly With this edition the Guide becomes available nationwide for the first time One excellent feature is the chapter on preparing writing samples a section of the MCAT often omitted or slighted in other guides Here is a step by step process for attacking writing sample prompts to The Stranger (MAXNotes Literature Study Guides) Kevin Kelly, 2013-01-01 REA s MAXnotes produce superior essays for Albert Camus The Stranger MAXnotes offer a fresh look at masterpieces of literature presented in a lively and interesting fashion Written by literary experts who currently teach the subject MAXnotes will enhance your understanding and enjoyment of the work MAXnotes are designed to stimulate independent thought about the literary work by raising various issues and thought provoking ideas and questions MAXnotes cover the essentials of what one should know about each work

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