

Orbitals and Bonding Test #12 SCH4U Name: _____

 K T C A
15 10 8 6

Assessment of Learning (Product)

Multiple Choice: Circle the most correct answer. (15K)

1. What is the electron configuration of potassium?

- A) $1s^2 2s^2 2p^2 3s^2 3p^2 4s^2$ C) $1s^2 2s^2 3s^2 3p^6 3d^2$
B) $1s^2 2s^2 2p^{10} 3s^2 3p^3$ D) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$

2. Which electron configuration represents a violation of the Pauli exclusion principle?



3. Which one of the following sets of quantum numbers can correctly represent a 3p orbital?

- | | | | | |
|------------|------------|------------|------------|------------|
| A) $n = 3$ | B) $n = 1$ | C) $n = 3$ | D) $n = 3$ | E) $n = 3$ |
| $l = 1$ | $l = 3$ | $l = 2$ | $l = 1$ | $l = 0$ |
| $m_l = 2$ | $m_l = 3$ | $m_l = 1$ | $m_l = -1$ | $m_l = 1$ |

4. How many electrons in an atom can have $n = 4$, $l = 2$?

- A) 6 B) 14 C) 10 D) 2 E) None of these

General Chemistry Ii Chapter Covalent Bonding Orbitals

Practice Test

Robert R. Redfield



General Chemistry II Chapter Covalent Bonding Orbitals Practice Test:

Unveiling the Magic of Words: A Report on "**General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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