

Factoring using GCF



Factorise the following expressions fully.

Section A

1) $8x + 24$

7) $21xy + 14x$

13) $9h^2g - 15h^3$

2) $15 + 25y$

8) $27ab - 18a^2$

14) $12c^2d^2 + 20c^3$

3) $32 - 40w$

9) $12s^2t + 28s$

15) $28a^3b^2 - 7a^2b$

4) $18c - 36$

10) $72wz + 45w^2z$

16) $60x^2y^3 - 35xy^2$

5) $16d^2 - 4d$

11) $22x^2y - 55xy^2$

17) $88s^4t + 56s^3t^2$

6) $12s + 60s^2$

12) $16k^3 + 24k^2$

18) $36p^3q^4 - 48p^4q^2$

Section B

1) $6 - 12gh + 3h$

7) $9wu - 27wuv + 45w$

13) $7xyz + xy^2 - x^2y$

2) $21st - 7t + 14$

8) $24gh - 12g + 15h$

14) $e^2f - 5e^3f^2 + e^2$

3) $22 - 44vw + 11v$

9) $132pqr - 96qr + 108pqrs$

15) $8st^2u - 32s^2t + 64st$

4) $4ab + 2b - abc$

10) $2x + xy - x^2$

16) $12g^3h - 9g^2h^2 + 18g^2h$

5) $5suv - 10sv + 15su$

11) $5k^2 - 10jk + k$

17) $\frac{1}{2}ab + \frac{3}{4}a^2 - a$

6) $16xy + 24y - 8xyz$

12) $9cd - 3c^2d + 12c$

18) $\frac{3}{4}x^4y - x^2y^3 + \frac{1}{2}x^3y^2$

Factoring Out Gcf Worksheet

Luciana C. de Oliveira, Marta Civil

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web lewis structures we also use lewis symbols to indicate the formation of covalent bonds which are shown in lewis structures drawings that describe the bonding in molecules and polyatomic ions for example when two chlorine atoms form a chlorine molecule they share one pair of electrons

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web this online quiz is intended to give you extra practice in identifying and drawing lewis dot structures as well as predicting ion formation this quiz aligns with the following ngss standard s hs ps1 1

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web jan 30 2023 a lewis structure is a very simplified representation of the valence shell electrons in a molecule it is used to show how the electrons are arranged around individual atoms in a molecule electrons are shown as dots or for bonding electrons as a line between the two atoms

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web jan 30 2023 follow these simple steps to draw lewis dot structures draw the atoms on paper and put dots around them to represent valence electrons of the atom be sure to have the correct number of electrons if the species is an ion add or subtract electrons corresponding to the charge of the ion

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