

Chemistry: EOC Review Packet

Chapter 1:

1.1 Matter and Change: Physical vs. Chemical Change (Properties that used questions)

Recall 1.1.1 Properties of Matter

- 1. **Physical properties** – characteristics of substances, specifically, characteristics observed without a changing into different substances
- 2. **Physical changes** – transformations involving only the change in state or form
- 3. **Chemical property** – characteristic that describes how substances interact with other substances to form new substances with different properties
- 4. **Chemical change** – chemical bonds are broken or created (transformations that form
- 5. **Elements** – pure substances made of only one type of atom
- 6. **Compounds** – put a substance with two or more elements combined into one substance
- 7. **Mixtures** – made of two or more
- 8. **Interactions** – substances combine differently
- 9. **Structure** 1.1.2 Molecular Structure

- 10. **Intermolecular forces** – particles interacting with one another and related energy in molecules or matter
- 11. **Intermolecular forces** – forces of attraction/repulsion between particles of matter

Changes	Physical Changes
State changes	Physical
Phase changes (Solid to Liquid or Gas)	Physical
Dissolving in water	Physical
Is transformation (Sugar to water)	Physical
Boiling	Physical
Freezing	Physical

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JL Elias



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