Chapter 9:	Electrons in	Atoms and	the !	Periodic Table

Date:

Section 1: Light and Quantized Energy - Notes

Objectives:

- Compare the wave and particle natures of light.
- Define a quantum of energy, and explain how it is related to an energy change of matter.
- Contrast continuous electromagnetic spectra and atomic emission spectra.

Unanswered Questions:

- In Rutherford's model, the atom's mass is concentrated in the ______

 and _____ move around it.
- This model doesn't explain how the ______ were arranged around the nucleus.
- This model also doesn't explain why ______ charged electrons aren't pulled into the ______ charged nucleus.

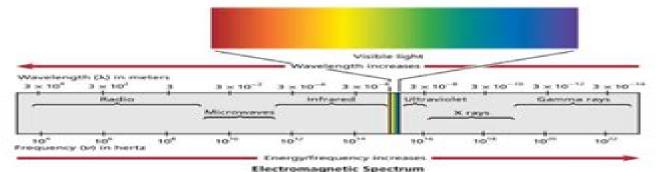
The Wave Nature of Light:

_____ when heated in a flame.

- Analysis of this light revealed that an element's ________ in its atoms.

 is related to the arrangement of the _______ in its atoms.

- The electromagnetic spectrum includes all forms of electromagnetic radiation:



Light And Quantized Energy Study Guide Answer

Ying Liu

Light And Quantized Energy Study Guide Answer:

Light And Quantized Energy Study Guide Answer Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the ability of words has are more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Light And Quantized Energy Study Guide Answer**, a literary masterpiece that delves deep to the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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