

7. Assuming all gas is removed from the tank when filling balloons, how many 0.75 L balloons can be filled from a tank that containing 375 g  $\text{CO}_2$  at 1.0 atm of pressure at a temperature of 298 K?

8. If a sample of  $\text{H}_2\text{S}$  gas at 1.00 atm in a 2.00 L container is allowed to expand into a 10.0 L container, what is the pressure in the container?

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