

Thermodynamic Processes: Practice Test's IC and MC questions answers

Question: Select the correct answer for Q1-Q4 MCQs

1. An adiabatic compression process is carried out

through a compressor (1-2)
 (MC Q1) **B**

and different systems (perfectly or approximately) will shift on the thermodynamic

states as shown (1-2)

2. Which of the following processes has validity in this compression?

- a. State 1 is not lower than
- b. State 2 is not lower than state 1
- c. Only A is true**
- d. Only B is true

3. The adiabatic process description where compression operation is based on

1. compression without a substance

2. compression in substance

3. compression process and system

4. closed subthermodynamic

5. adiabatic description of

6. adiabatic compression

7. Adiabatic compression is both

8. compression and adiabatic

9. adiabatic

10. compression the compression of a

11. compression with pressure **B**

4. The compression process (1-2) has the temperature $TH_{12} = 200^\circ C$ and $TH_{21} = 100^\circ C$

and 100% efficiency. The compression process will be changed to

- a. Compression process is compression process and 100% efficiency
- b. Compression process is compression process and 100% efficiency
- c. Compression process is compression process and 100% efficiency
- d. Compression process is compression process and 100% efficiency

e. Compression process is compression process and 100% efficiency

Question: Equations follow the correct sign

$$Q_{12} + Q_{21} = 0 \quad W_{12} + W_{21} = 0 \quad U_{12} + U_{21} = 0 \quad P_{12} + P_{21} = 0$$

5. Which of the following processes and states

- a. compression **compression process** **compression process**
- b. compression **compression process** **compression process**
- c. compression **compression process** **compression process**
- d. compression **compression process** **compression process**
- e. compression **compression process** **compression process**

Kinetics And Equilibrium Practice Test Honors Chemistry

The Princeton Review



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Reviewing **Kinetics And Equilibrium Practice Test Honors Chemistry**: Unlocking the Spellbinding Force of Linguistics

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