Be sure to answer all parts.

Give the sublevel designation, the allowable m_l values, and the number of orbitals for the following: n = 3, l = 1.

Sublevel designation:

- 0 5.
- 0 P
- 0 d
- 0 1
- 0 8
- 0 h

Allowable m, values:

- 0 6
- 0 -1, 0, +1
- 0 -2, -1, 0, +1, +2
- -3, -2, -1, 0, +1, +2, +3
- O -4, -3, -2, -1, 0, +1, +2, +3, +4
- -5, -4, -3, -2, -1, 0, +1, +2, +3, +4, +5

Number of orbitals:

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