

5. Complete the table below as shown by the example. Combine the ions into a correct formula, and name the compound.

|                  | $\text{CN}^-$                     | $\text{Cr}_2\text{O}_7^{2-}$ | $\text{P}^{3-}$ |
|------------------|-----------------------------------|------------------------------|-----------------|
| $\text{K}^+$     | $\text{KCN}$<br>potassium cyanide |                              |                 |
| $\text{Zn}^{2+}$ |                                   |                              |                 |
| $\text{Cr}^{3+}$ |                                   |                              |                 |

6. (optional) Complete the table below as shown by the example. Combine the ions into a correct formula, and name the compound.

|                | chlorate ion                         | sulfate ion | phosphate ion |
|----------------|--------------------------------------|-------------|---------------|
| cuprous ion    | $\text{CuClO}_3$<br>cuprous chlorate |             |               |
| nickel(II) ion |                                      |             |               |
| mercurous ion  |                                      |             |               |
| lead(IV) ion   |                                      |             |               |
| barium ion     |                                      |             |               |

# Laboratory Manual Prentice Hall Chemistry Answers

**Michael Brown**



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