

**COURSE STRUCTURE CLASS X**  
**(Annual Examination)**

**Marks: 80**

<b>Unit No.</b>	<b>Unit</b>	<b>Marks</b>	<b>Periods</b>
I	Chemical Substances-Nature and Behaviour	25	55
II	World of Living	23	50
III	Natural Phenomena	12	23
IV	Effects of Current	13	32
V	Natural Resources	07	20
	<b>Total</b>	<b>80</b>	
	<b>Internal assessment</b>	<b>20</b>	
	<b>Grand Total</b>	<b>100</b>	

**Theme: Materials**

**(55 Periods)**

**Unit I: Chemical Substances - Nature and Behaviour**

**Chemical reactions:**

Chemical equation, Balanced chemical equation, implications of a balanced chemical equation, types of chemical reactions: combination, decomposition, displacement, double displacement, precipitation, neutralization, oxidation and reduction.

**Acids, bases and salts:**

Their definitions in terms of furnishing of  $H^+$  and  $OH^-$  ions, General properties, examples and uses, concept of pH scale (Definition relating to logarithm not required), importance of pH in everyday life; preparation and uses of Sodium Hydroxide, Bleaching powder, Baking soda, Washing soda and Plaster of Paris.

**Metals and nonmetals:**

Properties of metals and non-metals; Reactivity series; Formation and properties of ionic compounds; Basic metallurgical processes; Corrosion and its prevention.

**Carbon compounds:**

Covalent bonding in carbon compounds. Versatile nature of carbon. Homologous series. Nomenclature of carbon compounds containing functional groups (halogens, alcohol, ketones, aldehydes, alkanes and alkynes), difference between saturated hydrocarbons and unsaturated hydrocarbons. Chemical properties of carbon compounds (combustion, oxidation, addition and substitution reaction). Ethanol and Ethanoic acid (only properties and uses), soaps and detergents.

**Periodic classification of elements:**

Need for classification, early attempts at classification of elements (Dobereiner's Triads, Newland's Law of Octaves, Mendeleev's Periodic Table). Modern periodic table, gradation in properties, valency, atomic number, metallic and non-metallic properties.

# Lab Manual Science Class 10 Cbse Syllabus

**JG Myers**



## **Lab Manual Science Class 10 Cbse Syllabus:**

## Unveiling the Power of Verbal Art: An Mental Sojourn through **Lab Manual Science Class 10 Cbse Syllabus**

In a world inundated with screens and the cacophony of quick communication, the profound energy and psychological resonance of verbal art usually disappear into obscurity, eclipsed by the continuous onslaught of sound and distractions. However, located within the musical pages of **Lab Manual Science Class 10 Cbse Syllabus**, a charming perform of fictional brilliance that impulses with fresh feelings, lies an unique journey waiting to be embarked upon. Published by way of a virtuoso wordsmith, this exciting opus instructions viewers on a psychological odyssey, delicately revealing the latent possible and profound affect stuck within the elaborate internet of language. Within the heart-wrenching expanse of this evocative analysis, we can embark upon an introspective exploration of the book is key themes, dissect their fascinating writing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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