# CHEMICAL KINETICS SAMPLE REPORT

#### Special Notes

- The report must be written in ink (blue pen), no pencil allowed. However, the graph can be drawn in pencil.
- The layout of the report must be tidy (no white out) and legible in writing.(Work out all the solutions on some scrap papers first, then copy to the report sheets.)
- Page 7 1
  - Record exact temperatures in °C to <u>1 decimal</u>.
  - ② Both the "Time" in seconds and "Relative Rate" must be integers.
  - Report the "Initial [I]", "Initial [S<sub>2</sub>O<sub>8</sub><sup>2</sup>]" and "Rate" in 3 s.f.
- Page 7 2
  - ② Show all calculations for:
    - (A) Concentration of initial [I"],
    - (B) Concentration of initial [S<sub>2</sub>O<sub>8</sub><sup>2-</sup>],
    - (C) Rate of Reaction of S<sub>2</sub>O<sub>x</sub><sup>2-</sup>, and
    - (D) Relative Rates
  - Keep all calculations in 3 s.f.. However, Relative Rate, Order of Reaction, and "m" and "n" must be reported as integers.
- Page 7 3
  - ① For Question I, the conclusion must include: rate ~ [I-].
  - To Proposition 2, the conclusion must include: rate =  $[S_2O_8^{2-}]$ .
  - $\odot$  For Question 3, the rate law is: rate =  $k[I^-][S_2O_8^{2-}]$
- Page 7 4
  - Show calculations of rate constants, k, for all five reactions.
  - ② The <u>units</u> of rate constant, k, is M<sup>-1</sup>-s<sup>-1</sup>.
  - ① The accuracy of k is between  $3.30 \times 10^{-3}$  to  $3.90 \times 10^{-3}$  M<sup>-1</sup>-s<sup>-1</sup> at 22-23°C.
  - The precision of k ±0.30 × 10⁻3 M⁻¹⁻s⁻¹ between highest and lowest k.

# **Kinetics Of A Chemical Reaction Lab Report**

United States. Environmental Protection Agency

**Kinetics Of A Chemical Reaction Lab Report:** 

Unveiling the Magic of Words: A Report on "Kinetics Of A Chemical Reaction Lab Report"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Kinetics Of A Chemical Reaction Lab Report**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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