

Ekurhuleni North	District Paper – Mathematics Paper 2 Grade 11 Marking Guideline	June 2023
SOLUTION NB: Accept valid alternate solutions. Consistent Accuracy applies to the whole memo.		
QUESTION 1		
1.1	$m_{HM} = \frac{6 - 4}{3 + 1}$ $= \frac{1}{2}$	✓ subs into formula ✓ answer
1.2	$R(x; y) = \left(\frac{3 + 4}{2}; \frac{6 + 1}{2} \right)$ $= \left(\frac{7}{2}; \frac{7}{2} \right)$	✓ Correct Substitution ✓ Answer
1.3	$m_{HM} = m_{AP}$ $\tan \theta = \frac{1}{2}$ $\theta = 26,57^\circ$	✓ $HM \parallel AT$ ✓ Correct substitution ✓ Answer
1.4	R is the midpoint of diagonal AH (Diags of a parm) Let the coordinates of A be $(x; y)$. $\frac{-1 + x}{2} = \frac{7}{2}$ $x = 8$ $\frac{4 + y}{2} = \frac{7}{2}$ $y = 3$ $\therefore A(8; 3)$	$\checkmark \frac{-1+x}{2} = \frac{7}{2}$ $\checkmark 8$ $\checkmark \frac{4+y}{2} = \frac{7}{2}$ $\checkmark 3$
1.5	$m_{AT} = \frac{1}{2} \text{ (parallel lines)}$ $y = \frac{1}{2}x + c$ Sub $T(4; 1)$	 ✓ Sub $T(4; 1)$ ✓ $c = -1$

Grade 11 Mathematics P2 Exemplar 20114 Memo

Christian G. Meyer



Grade 11 Mathematics P2 Exemplar 20114 Memo:

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web electrochemical cell 1 select the following red wire metal silver solution 1 0 m agno 3 aq black wire metal copper solution 1 0 m cu no 3 2 aq label the diagram below showing the following electrochemical cell that you built and observed label electrode with the metals show direction of electron flow show electrode polarity

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web choose 1 answer choice a at equilibrium e c e l l text cell cell start text end text start subscript c e l l end subscript calculates to a value of 0 and the battery is considered dead because q becomes k e q text eq eq

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web this is a post lab for electrochemistry determining an activity series using galvanic cells these are the first 6 questions and this is my data but i only need answers for 7 and 8 1 using copper as the standard cu cu cell potential 0 determine the potential for each of the reactions between two metals