

LIFE ORIENTATION

GRADE 11

NOTES

Grade 11 Lo Exemplar

Juan Eusebio Nieremberg



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 solid calcium carbonate is heated and decomposes to solid calcium oxide and carbon dioxide gas
 gaseous butane C_4H_{10} reacts with diatomic oxygen gas to yield gaseous carbon dioxide and water vapor

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$$\begin{aligned} & \text{Na} + 3\text{PO}_4 + 3\text{KOH} + 3\text{NaOH} + 1\text{K}_3\text{PO}_4 + 1\text{MgF}_2 + 1\text{Li}_2\text{CO}_3 + 1\text{MgCO}_3 + 2\text{LiF} + 1\text{P}_4 + 3\text{O}_2 + 2\text{P}_2 + 3\text{RbNO}_3 + 1\text{BeF}_2 + 1\text{BeNO}_3 + 2\text{RbF} \end{aligned}$$

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$$\begin{aligned} & \text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3 \\ & 2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2 \\ & 2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2 \\ & 2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O} \\ & 5\text{Pb} + \text{OH}^- + 2\text{HCl} + 2\text{H}_2\text{O} + 1\text{PbCl}_2 + 6\text{Al} + 3\text{Br}_2 + 3\text{K}_2\text{SO}_4 + 6\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O} + \text{C}_3\text{H}_8 + 5\text{O}_2 + 3\text{CO}_2 + 4\text{H}_2\text{O} \end{aligned}$$

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$$\begin{aligned} & 1\text{O}_2 + \text{NH}_3 + \text{HNO}_3 + \text{H}_2\text{O} + 2\text{O}_2 + 3\text{H}_2\text{O} + 2\text{H}_2\text{O} + 4\text{RbH}_2\text{O} + \text{RbOH} + \text{H}_2 + 5\text{Cl}_2 + \text{KBr} + \text{KCl} + \text{Br}_2 + 6\text{N}_2 + 2\text{H}_2 + \text{NH}_3 + 7\text{N}_2\text{F}_6 + 3\text{Ca} + \text{Nd} + \text{CaF}_2 + 8\text{CH}_4 + \text{H}_2\text{O} + \text{CO}_2 + \text{H}_2 + 9\text{C}_2\text{H}_5\text{OH} \end{aligned}$$

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