

# Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)

## 1. Physical Science Basics

- What tool(s) would you want to use to find the:
  - Mass of an object? triple beam balance Basic SI Unit? kg
  - Volume of an object? ruler or water displacement method Basic SI Unit? cm<sup>3</sup> or ml
  - Length of an object? Ruler Basic SI Unit? meter
  - Density of an object? graduated cylinder Basic SI Unit? ml
- What is the formula for calculating density? mass/volume
  - What is the density of a cube that is 25 g and has a volume of 3 cm<sup>3</sup>? 8g/cm<sup>3</sup>
  - Compare the density of solid H<sub>2</sub>O to liquid H<sub>2</sub>O, and explain what observable effect this has.  
solid H<sub>2</sub>O less dense than liquid H<sub>2</sub>O

## 2. Chemistry Basics

- Element (def.) substance made up of one kind of atom
- Atom (def.) basic building block of matter
- Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

## 3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

Scientist	Year	Major Discovery/ Contribution	Name of atomic model
John Dalton	1808	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.	Billiard Ball Model
J.J. Thomson	1897	Negatively charged electron	Plum Pudding Model
Niels Bohr	1913	Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell ( <u>energy level</u> ) to another.	Bohr Model
Ernest Rutherford	1909	1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus	Planetary Model

## 4. Periodic Table

- The current periodic table is arranged by increasing atomic number.
- Reading the Periodic Table
  - An element's properties can be predicted from its location in the periodic table.
  - The atomic number represents the number of protons.
  - The atomic mass represents the number of neutrons and protons.

# Final Exam Study Guide Physical Science

**N Colangelo**



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*X-kit Exam 2004 Physical Science*, 2005      **Physics Handbook & Study Guide Grade 11-12 IEB** Kevin Smith, 2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text      **Resources in Education**, 1998      *CHEMISTRY HANDBOOK & STUDY GUIDE Gr11-12 NE* Kevin Smith, 2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text      *Excel 2010 for Physical Sciences Statistics* Thomas J Quirk, Meghan Quirk, Howard Horton, 2013-07-10 This is the first book to show the capabilities of Microsoft Excel to teach physical sciences statistics effectively It is a step by step exercise driven guide for students and practitioners who need to master Excel to solve practical science problems If understanding statistics isn't your strongest suit you are not especially mathematically inclined or if you are wary of computers this is the right book for you Excel a widely available computer program for students and managers is also an effective teaching and learning tool for quantitative analyses in science courses Its powerful computational ability and graphical functions make learning statistics much easier than in years past However Excel 2010 for Physical Sciences Statistics A Guide to Solving Practical Problems is the first book to capitalize on these improvements by teaching students and managers how to apply Excel to statistical techniques necessary in their courses and work Each chapter explains statistical formulas and directs the reader to use Excel commands to solve specific easy to understand science problems Practice problems are provided at the end of each chapter with their solutions in an appendix Separately there is a full Practice Test with answers in an Appendix that allows readers to test what they have learned Includes 159 illustrations in color Suitable for undergraduates or graduate students      *Research in Education*, 1970      *Vibration Engineering Exam Study Guide* Cybellium, Welcome to the forefront of knowledge with Cybellium your trusted partner in mastering the cutting edge fields of IT Artificial Intelligence Cyber Security Business Economics and Science Designed for professionals students and enthusiasts alike our comprehensive books empower you to stay ahead in a rapidly evolving digital world Expert Insights Our books provide deep actionable insights that bridge the gap between theory and practical application Up to Date Content Stay current with the latest advancements trends and best practices in IT AI Cybersecurity Business Economics and Science Each guide is regularly updated to reflect the newest developments and challenges Comprehensive Coverage Whether you're a beginner or an advanced learner Cybellium books cover a wide range of topics from foundational principles to specialized knowledge tailored to your level of expertise Become part of a global network of learners and professionals who trust Cybellium to guide their educational journey [www.cybellium.com](http://www.cybellium.com)      *Guide to American Graduate Schools* Harold R. Doughty, 2009-02-24 For students planning further study after college the Guide to American Graduate Schools puts the

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first comprehensive quantitative analysis of intervention research in the learning disabilities field this volume synthesizes the results of 272 scientifically credible group and single subject studies in an effort to identify what works best for learning disabled children The book examines pertinent findings on all academic cognitive and behavioral domains Intervention outcomes are evaluated across instructional domains sample characteristics intervention parameters methodological

procedures and article characteristics Addressing such questions as the merits of inclusion settings and the relative benefits of direct and strategy instruction Swanson offers timely recommendations for instructional design assessment and policy

*Collateral Damage* Sharon L. Nichols, David C. Berliner, 2007-03-01 Drawing on their extensive research Nichols and Berliner document and categorize the ways that high stakes testing threatens the purposes and ideals of the American education system For more than a decade the debate over high stakes testing has dominated the field of education This passionate and provocative book provides a fresh perspective on the issue and powerful ammunition for opponents of high stakes tests Their analysis is grounded in the application of Campbell's Law which posits that the greater the social consequences associated with a quantitative indicator such as test scores the more likely it is that the indicator itself will become corrupted and the more likely it is that the use of the indicator will corrupt the social processes it was intended to monitor Nichols and Berliner illustrate both aspects of this corruption showing how the pressures of high stakes testing erode the validity of test scores and distort the integrity of the education system Their analysis provides a coherent and comprehensive intellectual framework for the wide ranging arguments against high stakes testing while putting a compelling human face on the data marshalled in support of those arguments

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**Who's the New Kid in Chemistry?** John D. Butler, 2013-12-12 Who's the New Kid in Chemistry offers an unprecedented look at student engagement and teacher best practices through the eyes of an educational

researcher enrolled as a public high school student Over the course of seventy nine consecutive days John D Butler participates in and observes Rhode Island 2013 Teacher of the Year Jessica M Waters s high school chemistry class documenting his experiences as they unfold Who s the New Kid in Chemistry is a compelling example of what can be accomplished when an educational researcher and teacher collaborate in the classroom This work includes a discussion on flexible homework assignments data driven instruction and thirty teacher best practices This book is an invaluable resource for teachers across all content areas masters and doctoral research method classes and future Teachers of the Year

**Graduate Education in the Chemical Sciences** National Research Council, Commission on Physical Sciences, Mathematics, and Applications, Board on Chemical Sciences and Technology, Chemical Sciences Roundtable, 2000-07-31 Graduate Education in the Chemical Sciences is a summary of the December 1999 workshop Graduate Education in the Chemical Sciences Issues for the 21st Century This workshop discussed the various features of graduate education in chemical science and technology Using case histories and their individual experiences speakers examined the current status of graduate education in the chemical sciences identified problems and opportunities and discussed possible strategies for improving the system The discussion was oriented toward the goal of generating graduates who are well prepared to advance the chemical sciences in academia government and industry in the next 5 to 10 years **The Science Teacher**, 1987

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