

Name: Daniela Martinez

Date: \_\_\_\_\_

## Student Exploration: Stoichiometry

**Vocabulary:** Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, molar mass, mole, molecular mass, stoichiometry

### Activity A: Moles

Get the Gizmo ready:

- Check that the equation is still:  
$$\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$$
- If not, click **New equation** until it reappears.



**Question:** How do scientists find the molecular mass and molar mass of a substance?

- Calculate:** The **molecular mass** of a molecule is the sum of the masses of each atom in the molecule. The unit of molecular mass is the universal mass unit (u).

Iron's atomic mass is 55.85 u, carbon's mass is 12.01 u, and oxygen's mass is 16.00 u.

- Calculate the molecular mass of carbon monoxide (CO) by adding the atomic mass of carbon and the atomic mass of oxygen: 28.01g
- Calculate the molecular mass of carbon dioxide (CO<sub>2</sub>): 44.01g
- Calculate the molecular mass of iron (III) oxide (Fe<sub>2</sub>O<sub>3</sub>): 159.7g

- Infer:** A mole of a substance has a mass in grams that is equal to the molecular mass. For example, a carbon atom has a mass of 12.01 u. A mole of carbon has a mass of 12.01 g. Based on their molecular masses, list the **molar mass** of each substance:

Fe<sub>2</sub>O<sub>3</sub> 159.7g   CO 28.01g   Fe 55.85g   CO<sub>2</sub> 44.01g

Check your answers on the Gizmo by inspecting the middle row of tiles on the right side of the Gizmo. These tiles show the units "1 mol" on top and "g" below.

- Practice:** Hydrogen has an atomic mass of 1.01 u. What is the molar mass of these substances?

H<sub>2</sub>O 18.02g   CH<sub>4</sub> 16.05g   H<sub>2</sub>CO<sub>3</sub> 62.03g   C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> 180.18g

# Exploration Sheet Answer Key Stoichiometry

**Richard Bailey**



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web nu nederlands 2f toets h1 lezen versie a noordhoff uitgevers bv 2014 pagina 1 van 3 toets lezen hoofdstuk 1 2f versie a ik herken onderwerp en hoofdgedachte van een tekst 1 1 ik herken schrijfdoele en tekstsoorten 1 2 ik controleer de betrouwbaarheid van informatie in een tekst 1 3 tekst 1

*het toetsen van begrijpend lezen nieuwsbegrip* - Dec 27 2021

web 1 bron nieuwsbegrip tekstentoets niveau b toets 1 versie 1 november meerkeuzetoetsen zijn in het nederlandse onderwijs waarschijnlijk de meest gebruikte toetsen om het begrijpend leesniveau van leerlingen te meten1 een bekend voorbeeld is de citotoets begrijpend lezen vaak bestaat de toets uit enkele teksten over

*nederlands havo vwo leerjaar 1 en 2* - Apr 30 2022

web feb 11 2016 1 5 februari 2016 hoofdstuk 3 lezen opdr 1 t m 3 spreken kijken en luisteren opdr 2 3 4 en 6 schrijven opdr 2 3 en 4 inleveren voor een cijfer maak de oefentoets op nieuw nederlands online 9 t m 12 feb hoofdstuk 3 woordenschat opdr 1 t m 8 grammatica zinsdelen opdr 2 t m 6 grammatica woordsoorten opdr 2 t m 7

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web oct 25 2012 nieuw nederlands samenvatting mavo havo vwo hoofdstuk 1 t m 6 tip in leerjaar 1 is nog weinig verschil mavo mag deze samenvatting ook gebruiken h 1 t m 6 van lezen 1 het onderwerp van een tekst 2 de hoofdgedachte 3 hoofd en bijzaken 4 zoekend lezen 5 feiten meningen en argumenten 6 tekstdoele 1 het

**antwoorden nederlands nieuw nederlands hoofdstuk 1 t m 5** - Sep 04 2022

web fictie hoofdstuk 1 opdracht 1 a 1 het rijmt geen leestekens alleen hoofdletters aan het begin van een couplet 2 3 strofen 3 twee teevee mond vond 4 een jongere wan het vraagt zich af hoe het later allemaal bij hem zou gaan 5 hij zag het op de tv

6 omdat hij wilde weten hoe het zou gaan b

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