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Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? *Frank can share his red marker with Flora and Flora can share her blue marker with Frank.*

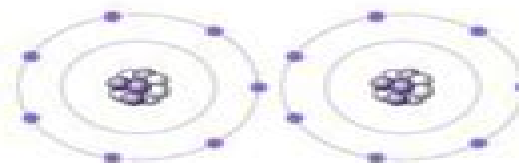
2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Olivia can share her purple and green markers with Otto and Otto can share his black and brown markers with Olivia.

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo™, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? *Each fluorine atom has seven valence electrons.*
2. Click **Pause** () . Drag an electron from the left atom to the right atom. Click **Play**. What happens? *The electrons that was dragged now orbits both atoms.*

Exploration Guide Covalent Bonds

N Noddings



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Table of Contents Exploration Guide Covalent Bonds

1. Understanding the eBook Exploration Guide Covalent Bonds
 - The Rise of Digital Reading Exploration Guide Covalent Bonds
 - Advantages of eBooks Over Traditional Books
2. Identifying Exploration Guide Covalent Bonds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Exploration Guide Covalent Bonds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Exploration Guide Covalent Bonds
 - Personalized Recommendations
 - Exploration Guide Covalent Bonds User Reviews and Ratings
 - Exploration Guide Covalent Bonds and Bestseller Lists
5. Accessing Exploration Guide Covalent Bonds Free and Paid eBooks
 - Exploration Guide Covalent Bonds Public Domain eBooks
 - Exploration Guide Covalent Bonds eBook Subscription Services
 - Exploration Guide Covalent Bonds Budget-Friendly Options
6. Navigating Exploration Guide Covalent Bonds eBook Formats

- ePub, PDF, MOBI, and More
- Exploration Guide Covalent Bonds Compatibility with Devices
- Exploration Guide Covalent Bonds Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Exploration Guide Covalent Bonds
 - Highlighting and Note-Taking Exploration Guide Covalent Bonds
 - Interactive Elements Exploration Guide Covalent Bonds
- 8. Staying Engaged with Exploration Guide Covalent Bonds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Exploration Guide Covalent Bonds
- 9. Balancing eBooks and Physical Books Exploration Guide Covalent Bonds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Exploration Guide Covalent Bonds
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Exploration Guide Covalent Bonds
 - Setting Reading Goals Exploration Guide Covalent Bonds
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Exploration Guide Covalent Bonds
 - Fact-Checking eBook Content of Exploration Guide Covalent Bonds
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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