

# PERIODIC TABLE

|   |  |                             |  |                                   |  |                             |  |                                |  |                              |  |                              |  |                                |  |                              |  |                              |  |                                 |  |  |  |                                 |  |  |  |   |  |  |  |   |  |                                |  |                                |  |                             |  |                                 |  |                              |  |                                |  |
|---|--|-----------------------------|--|-----------------------------------|--|-----------------------------|--|--------------------------------|--|------------------------------|--|------------------------------|--|--------------------------------|--|------------------------------|--|------------------------------|--|---------------------------------|--|--|--|---------------------------------|--|--|--|---|--|--|--|---|--|--------------------------------|--|--------------------------------|--|-----------------------------|--|---------------------------------|--|------------------------------|--|--------------------------------|--|
| 1 METALS  |  |                             |  |                                   |  |                             |  |                                |  |                              |  |                              |  |                                |  |                              |  | NONMETALS                    |  |                                 |  |  |  |                                 |  |  |  | 18 Noble Gases                          |  |  |  |   |  |                                |  |                                |  |                             |  |                                 |  |                              |  |                                |  |
| <div>1<br/>H<br/>Hydrogen<br/>1</div> <div>Atomic Number = Number of Protons = Number of Electrons<br/>Symbol<br/>Element Name<br/>Atomic Mass = Number of Protons + Number of Neutrons</div> |  |                             |  |                                   |  |                             |  |                                |  |                              |  |                              |  |                                |  |                              |  |                              |  |                                 |  |  |  |                                 |  |  |  | <div>2<br/>He<br/>Helium<br/>4</div>    |  |  |  |   |  |                                |  |                                |  |                             |  |                                 |  |                              |  |                                |  |
| 2<br>Li<br>Lithium<br>7   |  | 4<br>Be<br>Beryllium<br>9   |  | Transition Elements               |  |                             |  |                                |  |                              |  |                              |  |                                |  | 5<br>B<br>Boron<br>11        |  | 6<br>C<br>Carbon<br>12       |  | 7<br>N<br>Nitrogen<br>14        |  | 8<br>O<br>Oxygen<br>16                 |  | 9<br>F<br>Fluorine<br>19        |  | 10<br>Ne<br>Neon<br>20                   |  |   |  |  |  |   |  |                                |  |                                |  |                             |  |                                 |  |                              |  |                                |  |
| 3<br>Na<br>Sodium<br>23   |  | 12<br>Mg<br>Magnesium<br>24 |  |                                   |  |                             |  |                                |  |                              |  |                              |  |                                |  | 13<br>Al<br>Aluminum<br>27   |  | 14<br>Si<br>Silicon<br>28    |  | 15<br>P<br>Phosphorus<br>31     |  | 16<br>S<br>Sulfur<br>32                |  | 17<br>Cl<br>Chlorine<br>35      |  | 18<br>Ar<br>Argon<br>40                  |  |   |  |  |  |   |  |                                |  |                                |  |                             |  |                                 |  |                              |  |                                |  |
| 19<br>K<br>Potassium<br>39  |  | 20<br>Ca<br>Calcium<br>40   |  | 21<br>Sc<br>Scandium<br>45        |  | 22<br>Ti<br>Titanium<br>48  |  | 23<br>V<br>Vanadium<br>51      |  | 24<br>Cr<br>Chromium<br>52   |  | 25<br>Mn<br>Manganese<br>55  |  | 26<br>Fe<br>Iron<br>56         |  | 27<br>Co<br>Cobalt<br>59     |  | 28<br>Ni<br>Nickel<br>59     |  | 29<br>Cu<br>Copper<br>64        |  | 30<br>Zn<br>Zinc<br>65                 |  | 31<br>Ga<br>Gallium<br>70       |  | 32<br>Ge<br>Germanium<br>73              |  | 33<br>As<br>Arsenic<br>75               |  | 34<br>Se<br>Selenium<br>79               |  | 35<br>Br<br>Bromine<br>80               |  | 36<br>Kr<br>Krypton<br>84      |  |                                |  |                             |  |                                 |  |                              |  |                                |  |
| 37<br>Rb<br>Rubidium<br>85  |  | 38<br>Sr<br>Strontium<br>88 |  | 39<br>Y<br>Yttrium<br>89          |  | 40<br>Zr<br>Zirconium<br>91 |  | 41<br>Nb<br>Niobium<br>93      |  | 42<br>Mo<br>Molybdenum<br>96 |  | 43<br>Tc<br>Technetium<br>98 |  | 44<br>Ru<br>Ruthenium<br>101   |  | 45<br>Rh<br>Rhodium<br>103   |  | 46<br>Pd<br>Palladium<br>106 |  | 47<br>Ag<br>Silver<br>108       |  | 48<br>Cd<br>Cadmium<br>112             |  | 49<br>In<br>Indium<br>115       |  | 50<br>Sn<br>Tin<br>119                   |  | 51<br>Sb<br>Antimony<br>122             |  | 52<br>Te<br>Tellurium<br>128             |  | 53<br>I<br>Iodine<br>127                |  | 54<br>Xe<br>Xenon<br>131       |  |                                |  |                             |  |                                 |  |                              |  |                                |  |
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# Elements And The Periodic Table Pretest

**Brian W. Stewart**



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